

chemistry basics UNIT 1

created on:
9.25.21

LAB EQUIPMENT



beaker



erlenmeyer flask



test tube



graduated cylinder



crucible



laboratory burner



thermo-meter



pipette

DENSITY

$$d = \frac{m}{v}$$

density = mass / volume
↳ g/mL, g/cm³

- cross multiply if density is given
- mL = cm³ for volume

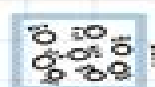
PARTICLE DIAGRAMS



Solids



liquids



gases

most dense → least dense

Celsius °C + Kelvin
+ 273
to make
Celsius °C
Kelvin

MEASURING



= 1.10 cm

this ruler measures to the hundredths place.

↳ always round 1 more place than shown on the measuring tool

- ALWAYS view measurements from eye level
- read from the meniscus (graduated cylinder)



CONVERSIONS

- use the conversions given to find the new conversions

$$\frac{1000 \text{ mL}}{1} \times \frac{1 \text{ L}}{1000 \text{ mL}} = \frac{1000}{1000} \quad 1000 \text{ mL} = 1 \text{ liter}$$

K H D B d c m — micro — nano — pico
kilo hecto deca base deci centi milli
bigger → smaller

* if you're converting left to right, the conversion will be smaller than the original #

significant figures

RULES:

- 0.0089040
- 340
- 340.
- 37890.00
- 0.06
- 1000

- all 0's before don't count, all after natural #'s & the decimal do.
- 0's before the tenths place will only count if there's a decimal after it.
- The 0 here counts because there is a decimal after it.
- If there are natural numbers before the decimal, anything after it counts.
- If there are no natural numbers before the decimal, sig figs start at the first natural # after the decimal.
- no decimal point after the 0's so only natural #'s count.

SCIENTIFIC NOTATION

- negative exponent = move decimal ←
↳ number will be smaller out of notation
- positive exponent = move decimal →
↳ number will be bigger out of notation

PERCENT ERROR

$$\% \text{ error} = \frac{|\text{accepted value} - \text{experimental}|}{\text{accepted value}} \times 100$$

* always include units in answers!

CHEMICAL CHANGE/PROPERTY

- change in atom formation
- underg. bond breaking / making to form something new
- anything having to do with reactions / reactivity



ex. flammability, Na is explosive in water

PHYSICAL PROPERTIES

- quality of matter that can be observed or measured without changing the bonds.
- no reactions, only things you can observe or see, or data given.



ex. the density of water is 1 g/mL, H₂O is clear.

Chemistry Of Life Note Taking Guide

Yan Bai



Chemistry Of Life Note Taking Guide:

Inclusion Strategies That Work for Adolescent Learners! Toby J. Karten, 2009-03-17 Help adolescent learners in inclusive environments develop strong academic emotional and social skills for successful post secondary options with lesson plans that focus on success and maximize students strengths interests and potentials Inclusion Strategies That Work for Adolescent Learners provides teachers with guidelines for scheduling preparation student educator responsibilities family community supports appropriate accommodations literacy and numeration strategies interdisciplinary approaches and more The author investigates inclusive strategies through reflective activities and pragmatic classroom applications that highlight productivity with evidence based practices for all learners *Focus on Physical Science California Edition* Michael J. Padilla, 2007 **Resources in Education** ,2000 **Cassell's household guide to every department of practical life** Anonymous, 2020-09-22 Reprint of the original first published in 1869 *Manual and Course of Study for the Public Schools of Texas, 1915* Texas. State Department of Education, 1915 **Life Cycle Assessment Handbook** Mary Ann Curran, 2012-11-07 The first book of its kind the Life Cycle Assessment Handbook A Guide for Environmentally Sustainable Products will become an invaluable resource for environmentally progressive manufacturers and suppliers product and process designers executives and managers and government officials who want to learn about this essential component of environmental sustainability As the last several decades have seen a dramatic rise in the application of Life Cycle Assessment LCA in decision making the interest in the life cycle concept as an environmental management and sustainability tool continues to grow The LCA Handbook offers a look at the role that life cycle information in the hands of companies governments and consumers may have in improving the environmental performance of products and technologies It concisely and clearly presents the various aspects of LCA in order to help the reader better understand the subject The content of the book was designed with a certain flow in mind After a high level overview to describe current views and state of the practice of LCA it presents chapters that address specific LCA methodological issues including creating life cycle inventory life cycle impact assessment and capturing eco systems services These are followed by example applications of LCA in the agri food industry sustainable supply chain management solid waste management mining and mineral extraction forest products buildings product innovation and sustainable chemistry and engineering The international success of the sustainability paradigm needs the participation of many stakeholders including citizens corporations academia and NGOs The handbook links LCA and responsible decision making and how the life cycle concept is a critical element in environmental sustainability It covers issues such as building capacity in developing countries and emerging economies so that they are more capable of harnessing the potential in LCA for sustainable development Governments play a very important role with the leverage they have through procurement regulation international treaties tax incentives public outreach and other policy tools This compilation points to the clear trend for incorporating life cycle information into the design and development processes for

products and policies just as quality and safety concerns are now addressed throughout product design and development

Technical Note, 1913 Experiment Station Record United States. Office of Experiment Stations, 1913 *Guide to Literature of Home and Family Life* Annie Robertson Dyer, 1924 A Manual of Antropometry; Or, A Guide to the Physical Examination and Measurement of the Human Body Charles Roberts, 1878

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