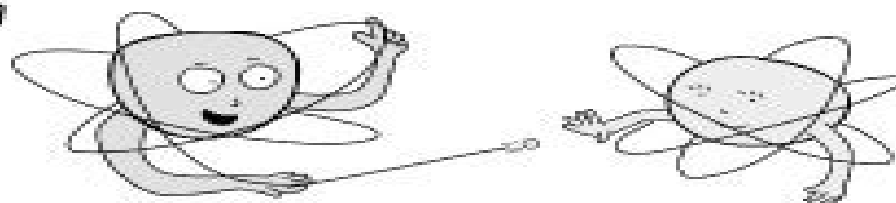


Chapter 11: Oxidations and Reductions Charge the World

Two Questions That Should Be Easy

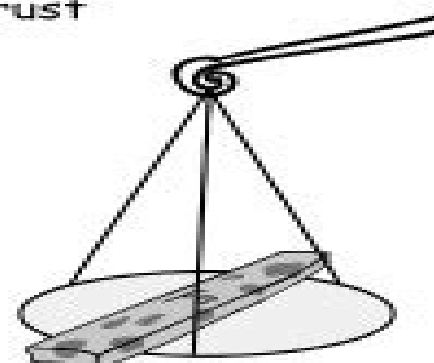
1. Under what conditions does an oxidizing agent donate electrons?

- a) Only at very high temperatures.
- b) During electrolysis.
- c) When a reducing agent is present.
- d) By definition, a chemical behaves as an oxidizing agent only when it accepts electrons.



2. What would happen to the weight of an iron bar when the iron rusts? Assume that the rust does not fall off the bar.

- a) The weight of the bar would increase because it is gaining oxygen.
- b) The weight would be the same because matter is neither created nor destroyed in a chemical reaction.
- c) The weight would be the same because the iron is only being oxidized.
- d) The weight would decrease because iron is being lost as iron oxide forms.
- e) The weight would decrease because iron oxide is less dense than iron.



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Dr. S C Rastogi, ,Er. Meera Goyal



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