

Lesson Overview

2.3 Carbon Compounds

THINK ABOUT IT

In the early 1800s, many chemists called the compounds created by organisms “organic,” believing they were fundamentally different from compounds in nonliving things.

We now understand that the principles governing the chemistry of living and nonliving things are the same, but the term “organic chemistry” is still around.

Today, organic chemistry means the study of compounds that contain bonds between carbon atoms, while inorganic chemistry is the study of all other compounds.

The Chemistry of Carbon

What elements does carbon bond with to make up life’s molecules?

The Chemistry of Carbon

What elements does carbon bond with to make up life’s molecules?

Carbon can bond with many elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen to form the molecules of life.

The Chemistry of Carbon

Carbon atoms have four valence electrons, allowing them to form strong covalent bonds with many other elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen.

Living organisms are made up of molecules that consist of carbon and these other elements.

The Chemistry of Carbon

Carbon atoms can also bond to each other, which gives carbon the ability to form millions of different large and complex structures.

Carbon-carbon bonds can be single, double, or triple covalent bonds.

Chains of carbon atoms can even close up on themselves to form rings.

Macromolecules

What are the functions of each of the four groups of macromolecules?

Macromolecules

What are the functions of each of the four groups of macromolecules?

Living things use carbohydrates as their main source of energy. Plants, some animals, and other organisms also use carbohydrates for structural purposes.

Concept Review Compounds Of Carbon Answers

TD Snyder



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