

Concentration and Molarity II: Dilution and Evaporation PhET-Chemistry Labs

Introduction: In the first part of this lab, you learned about the actions of solid salts added to water to form solutions of various concentrations. In this exercise, you will use a powerful and simple formula to determine the concentration of a solution when it has been diluted with additional water or made more concentrated due to evaporation of water.



Concentration

Some handy vocabulary for you to define:

Molarity _____

Dilution _____

Evaporation _____

Procedure: PhET → Play with the Sims → Chemistry → Concentration Main Menu

$$M_1V_1 = M_2V_2$$

Part I: Dilution

At this point you should be familiar with the Concentration simulation. Because some time may have passed, take a few minutes to re-learn the simulations. Pay particular attention to the effect of evaporation and addition of water to the solution's concentration.



How does the concentration change as additional water is added? _____

Why? _____



How does the concentration change as evaporation occurs? _____

Why is this? _____

Does evaporation change the concentration of a saturated solution? _____

Why is this? _____



Using the concentrated solution spigot, add a ½ Liter of Drink Mix to an empty beaker. What is the concentration? _____

_____ Is this solution saturated? _____ How do you know? _____

Fill the beaker with another ½ Liter of water. What is the new concentration? _____

Complete the table below, using Cobalt chloride in an empty beaker, writing the concentration in the boxes provided.

Only .25L of spigot solution	.25L spigot + .25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

Repeat the exercise, using Potassium dichromate in an empty beaker.

Only .25L of spigot solution	.25L spigot + .25L water	.25L spigot + .50 L water	.25L spigot + .75 L water

What do you notice about the concentration change as each addition of .25L of water is added to the concentrated spigot solution? _____

Concentration And Molarity Ii Dilution And Evaporation

Fred Leighton



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Table of Contents Concentration And Molarity Ii Dilution And Evaporation

1. Understanding the eBook Concentration And Molarity Ii Dilution And Evaporation
 - The Rise of Digital Reading Concentration And Molarity Ii Dilution And Evaporation
 - Advantages of eBooks Over Traditional Books
2. Identifying Concentration And Molarity Ii Dilution And Evaporation
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Concentration And Molarity Ii Dilution And Evaporation
 - User-Friendly Interface
4. Exploring eBook Recommendations from Concentration And Molarity Ii Dilution And Evaporation
 - Personalized Recommendations
 - Concentration And Molarity Ii Dilution And Evaporation User Reviews and Ratings

- Concentration And Molarity Ii Dilution And Evaporation and Bestseller Lists
- 5. Accessing Concentration And Molarity Ii Dilution And Evaporation Free and Paid eBooks
 - Concentration And Molarity Ii Dilution And Evaporation Public Domain eBooks
 - Concentration And Molarity Ii Dilution And Evaporation eBook Subscription Services
 - Concentration And Molarity Ii Dilution And Evaporation Budget-Friendly Options
- 6. Navigating Concentration And Molarity Ii Dilution And Evaporation eBook Formats
 - ePub, PDF, MOBI, and More
 - Concentration And Molarity Ii Dilution And Evaporation Compatibility with Devices
 - Concentration And Molarity Ii Dilution And Evaporation Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Concentration And Molarity Ii Dilution And Evaporation
 - Highlighting and Note-Taking Concentration And Molarity Ii Dilution And Evaporation
 - Interactive Elements Concentration And Molarity Ii Dilution And Evaporation
- 8. Staying Engaged with Concentration And Molarity Ii Dilution And Evaporation
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Concentration And Molarity Ii Dilution And Evaporation
- 9. Balancing eBooks and Physical Books Concentration And Molarity Ii Dilution And Evaporation
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Concentration And Molarity Ii Dilution And Evaporation
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Concentration And Molarity Ii Dilution And Evaporation
 - Setting Reading Goals Concentration And Molarity Ii Dilution And Evaporation
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Concentration And Molarity Ii Dilution And Evaporation
 - Fact-Checking eBook Content of Concentration And Molarity Ii Dilution And Evaporation
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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