

GENERAL CHEMISTRY REFERENCE SHEET

Common Metric Prefixes and Units		
Unit	Symbol	Conversions
meter	m	1 m = 39.37 inches
centimeter	cm	1 inch = 2.54 cm exactly
Angstrom	Å	1 Å = 1×10^{-10} m
gram	g	1 g = 0.002205 lb
liter	L	1 L = 1000 cm ³ = 1.056710 qt
Kelvin	K	K = °C + 273.15
kilo	k	10 ³
deci	d	10 ⁻¹
centi	c	10 ⁻²
milli	m	10 ⁻³
micro	μ	10 ⁻⁶
nano	n	10 ⁻⁹
pico	p	10 ⁻¹²

Nomenclature

Common Cations		
Charge	Formula	Name
1+	H ⁺	hydrogen ion
	H ₃ O ⁺	hydronium ion
	Alkali metal ⁺	alkali metal ion
	NH ₄ ⁺	ammonium ion
	Ag ⁺	silver ion
2+	Cu ⁺	copper(I)
	Alkaline earth metal ²⁺	alkaline earth metal ion
	Zn ²⁺	zinc ion
	Co ²⁺	cobalt(II)
	Cu ²⁺	copper(II)
	Fe ²⁺	iron(II)
	Mn ²⁺	manganese(II)
	Hg ₂ ²⁺	mercury(I)
	Hg ₂ ²⁺	mercury(II)
	Ni ²⁺	nickel(II)
	Pb ²⁺	lead(II)
	Sn ²⁺	tin(II)
3+	Al ³⁺	aluminum ion
	Cr ³⁺	chromium(III)
	Fe ³⁺	iron(III)

Common Anions		
Charge	Formula	Name
1-	H ⁻	hydride ion
	F ⁻	fluoride ion
	Cl ⁻	chloride ion
	Br ⁻	bromide ion
	I ⁻	iodide ion
	CN ⁻	cyanide ion
	OH ⁻	hydroxide ion
	NO ₃ ⁻	nitrate
	NO ₂ ⁻	nitrite
	HCO ₃ ⁻	hydrogen carbonate or bicarbonate ion
	HSO ₄ ⁻	hydrogen sulfate ion

Charge	Formula	Name
1-	ClO ₄ ⁻	perchlorate ion
	ClO ₃ ⁻	chlorate ion
	ClO ₂ ⁻	chlorite ion
	ClO ⁻	hypochlorite ion
	MnO ₄ ⁻	permanganate ion
	H ₂ PO ₄ ⁻	dihydrogen phosphate ion
	C ₂ H ₃ O ₂ ⁻ (or CH ₃ COO ⁻)	acetate ion
2-	O ²⁻	oxide ion
	S ²⁻	sulfide ion
	SO ₄ ²⁻	sulfate ion
	SO ₃ ²⁻	sulfite ion
	CO ₃ ²⁻	carbonate ion
	Cr ₂ O ₇ ²⁻	dichromate ion
3-	HPO ₄ ²⁻	hydrogen phosphate
	PO ₄ ³⁻	phosphate ion

Other Compounds

Formula	Name
H ₂ O	water
NH ₃	ammonia
CH ₄	methane (natural gas)
O ₃	ozone
H ₂ O ₂	hydrogen peroxide
H ₂ , N ₂ , O ₂ , F ₂ , Cl ₂ , Br ₂ , I ₂	diatomic molecules
He, Ne, Ar	Noble gases

Acids

Formula	Name
HNO ₃	nitric acid (strong)
H ₂ SO ₄	sulfuric acid (strong)
HCl	hydrochloric acid (strong)
HBr	hydrobromic acid (strong)
HI	hydroiodic acid (strong)
HClO ₄	perchloric acid (strong)
H ₃ PO ₄	phosphoric acid (weak)
H ₂ CO ₃	carbonic acid (weak)
HC ₂ H ₃ O ₂ (or CH ₃ COOH)	acetic acid (weak)
HF	hydrofluoric acid

Bases

Formula	Name
NaOH	sodium hydroxide (strong)
KOH	potassium hydroxide (strong)
Mg(OH) ₂	magnesium hydroxide (strong)
Ca(OH) ₂	calcium hydroxide (strong)
NH ₃	ammonia (weak)
NR ₃	other amines (weak)

Cst Chemistry Reference Sheet

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