

CSP Exam Practice Test 2

Q1. In the workplace, which specific function of the skin holds particular relevance for safeguarding against exposure to hazardous materials in the context of safety assessments?

- A. Sensory perception
- B. Immune function
- C. Fluid balance
- D. Regulation of internal processes

Q2. Elucidate the relationship between atomic number, electron configuration, and chemical reactivity in the formation of compounds, considering the periodic table.

- A. Atomic number directly correlates with chemical reactivity.
- B. Electron configuration is unrelated to atomic number.
- C. Chemical reactivity is solely determined by electron configuration.
- D. Atomic number and electron configuration collectively influence chemical reactivity.

Q3. How do specific factors relate to the electronegativity of the atoms involved and how do these specific factors influence the strength of a covalent bond in chemical bonding?

- A. Molecular weight and atomic radius; inversely proportional
- B. Electronegativity difference and number of shared electrons; directly proportional
- C. Ionization energy and atomic mass; inversely proportional
- D. Atomic number and bond length; directly proportional

Q4. How does the ionized nature of plasma, as the fourth state of matter pose unique

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A Loxley



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