

Enzyme and Temperature

Background

For the experiment an enzyme (catalase) breaking down hydrogen peroxide when heated will be used and tested.

Introduction

Enzymes are proteins that increase the rate of a chemical reaction (catalysis) without being consumed themselves. As catalysts, enzymes are not changed or consumed during the reaction. They are highly specific, meaning they only work on one type of substrate. Enzymes are also sensitive to temperature, pH, and other factors.

Key Concepts

Enzymes are biological catalysts that speed up chemical reactions.

Variables

Independent	Dependent	Controlled
Temperature	Rate of reaction (measured by volume of oxygen gas produced)	Amount of substrate (hydrogen peroxide), amount of enzyme (catalase), pH of solution

Procedure

Step 1	Step 2	Step 3
		
Preparation of hydrogen peroxide solution	Addition of catalase enzyme	Measurement of oxygen gas production

1. Preparation of hydrogen peroxide solution: 10 mL of 3% hydrogen peroxide solution is prepared in a test tube.

2. Addition of catalase enzyme: 1 mL of catalase enzyme solution is added to the test tube.

3. Measurement of oxygen gas production: The volume of oxygen gas produced is measured using a gas syringe.

Results

Time (min)	Volume of oxygen gas produced (mL)
0	0
1	1.5
2	3.0
3	4.5
4	6.0
5	7.5
6	9.0
7	10.5
8	12.0
9	13.5
10	15.0

Discussion

Time (min)	Volume of oxygen gas produced (mL)
0	0
1	1.5
2	3.0
3	4.5
4	6.0
5	7.5
6	9.0
7	10.5
8	12.0
9	13.5
10	15.0

Enzymes are highly specific and sensitive to temperature. The rate of reaction increases with temperature up to an optimal point, after which it decreases.



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Table of Contents Ccbc Biol 110 Lab Manual Enzymes Results

1. Understanding the eBook Ccbc Biol 110 Lab Manual Enzymes Results
 - The Rise of Digital Reading Ccbc Biol 110 Lab Manual Enzymes Results
 - Advantages of eBooks Over Traditional Books
2. Identifying Ccbc Biol 110 Lab Manual Enzymes Results
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ccbc Biol 110 Lab Manual Enzymes Results
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ccbc Biol 110 Lab Manual Enzymes Results
 - Personalized Recommendations
 - Ccbc Biol 110 Lab Manual Enzymes Results User Reviews and Ratings
 - Ccbc Biol 110 Lab Manual Enzymes Results and Bestseller Lists

5. Accessing Ccbc Biol 110 Lab Manual Enzymes Results Free and Paid eBooks
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 - Ccbc Biol 110 Lab Manual Enzymes Results Budget-Friendly Options
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 - Ccbc Biol 110 Lab Manual Enzymes Results Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ccbc Biol 110 Lab Manual Enzymes Results
 - Highlighting and Note-Taking Ccbc Biol 110 Lab Manual Enzymes Results
 - Interactive Elements Ccbc Biol 110 Lab Manual Enzymes Results
8. Staying Engaged with Ccbc Biol 110 Lab Manual Enzymes Results
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ccbc Biol 110 Lab Manual Enzymes Results
9. Balancing eBooks and Physical Books Ccbc Biol 110 Lab Manual Enzymes Results
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ccbc Biol 110 Lab Manual Enzymes Results
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ccbc Biol 110 Lab Manual Enzymes Results
 - Setting Reading Goals Ccbc Biol 110 Lab Manual Enzymes Results
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ccbc Biol 110 Lab Manual Enzymes Results
 - Fact-Checking eBook Content of Ccbc Biol 110 Lab Manual Enzymes Results
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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