

CHEMISTRY

UNITS, SCIENTIFIC NOTATION, AND SIGNIFICANT FIGURES

DEFINING UNITS

When trying to describe the quantity or quality of something, whether a chemical substance or physical phenomenon, it is helpful to have some standard measure to refer to. A unit is nothing more than a standard by which a measured value can be described. For instance, in the old English system, a foot was just that—the length of a man's foot. This was helpful in that when a distance between, say, a house and a water well was described in feet, one who had never seen the dwelling in question would have an instant idea of how far away the well was, based solely on a verbal description. The concept today hasn't changed, although our standards make better scientific sense and are more exact.

SYSTEMS OF UNITS

There are two different sets of units used in scientific measurements: British Engineering units and the metric system (also called the Standard International (SI) system). Both are based on standards, though those underlying the British system can be shockingly inscrutable. The metric system is based on a far less arbitrary system of tens, and is far more rational, intelligent, and easier to use. In general, people tend to stick with the system with which they are most familiar, and have the most intuitive feeling for. The United States uses the British system; most of the rest of the world (including Britain) uses the metric system. In terms of science, the metric system is almost always used. For Americans, it is therefore often necessary to make conversions between the two. Conversion tables reside in the "Reference" pull-down menu at the top of most SparkNotes pages.

METRIC UNITS (SI UNITS)

The standards metric terms for commonly measured characteristics are listed in below.

Property	Unit	Symbol
Mass	kilograms	kg
Length	meters	m
Volume	liters	L
Time	seconds	s
Electric Current	ampere	A
Temperature	Kelvin, Celsius	K, C
Intensity of light	candela	cd
Amount of a Substance	mole	mol

Temperature in the table above is expressed in both Kelvin (K) and Celsius (C). One kelvin is exactly the same as one C, except that the Kelvin scale starts at absolute zero, or -273.15°C , the lowest temperature possible, instead of the freezing point of water. Therefore water freezes at 273.15 K or 0°C .

PREFIXES FOR UNITS

In addition to the basic metric units described above, there are prefixes to indicate larger or smaller quantities. For example, a meter refers to a standard metric measure of length. A millimeter refers to a measure that is one thousandth the size of a meter (one thousand millimeters fit into a meter); a kilometer refers to a distance one thousand times longer than a meter (one thousand meters fit into a kilometer). By wisely using these prefixes, you can avoid having to use huge numbers or having to resort to scientific notation. The table below shows the most commonly encountered prefixes:

Prefix	Symbol	Value	Description
pico	p	10^{-12}	1 picoliter, (pL) = 0.000000000001 L
nano	n	10^{-9}	1 nanogram, (ng) = 0.000000001 g



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Table of Contents Chemistry Speedy Study Guides Speedy Publishing

1. Understanding the eBook Chemistry Speedy Study Guides Speedy Publishing
 - The Rise of Digital Reading Chemistry Speedy Study Guides Speedy Publishing
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Speedy Study Guides Speedy Publishing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Chemistry Speedy Study Guides Speedy Publishing
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Speedy Study Guides Speedy Publishing
 - Personalized Recommendations
 - Chemistry Speedy Study Guides Speedy Publishing User Reviews and Ratings
 - Chemistry Speedy Study Guides Speedy Publishing and Bestseller Lists

5. Accessing Chemistry Speedy Study Guides Speedy Publishing Free and Paid eBooks
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 - Chemistry Speedy Study Guides Speedy Publishing eBook Subscription Services
 - Chemistry Speedy Study Guides Speedy Publishing Budget-Friendly Options
6. Navigating Chemistry Speedy Study Guides Speedy Publishing eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Speedy Study Guides Speedy Publishing Compatibility with Devices
 - Chemistry Speedy Study Guides Speedy Publishing Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Speedy Study Guides Speedy Publishing
 - Highlighting and Note-Taking Chemistry Speedy Study Guides Speedy Publishing
 - Interactive Elements Chemistry Speedy Study Guides Speedy Publishing
8. Staying Engaged with Chemistry Speedy Study Guides Speedy Publishing
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Speedy Study Guides Speedy Publishing
9. Balancing eBooks and Physical Books Chemistry Speedy Study Guides Speedy Publishing
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Speedy Study Guides Speedy Publishing
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Speedy Study Guides Speedy Publishing
 - Setting Reading Goals Chemistry Speedy Study Guides Speedy Publishing
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Speedy Study Guides Speedy Publishing
 - Fact-Checking eBook Content of Chemistry Speedy Study Guides Speedy Publishing
 - Distinguishing Credible Sources
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- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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