CHEMISTRY EQUATIONS & ANSWERS

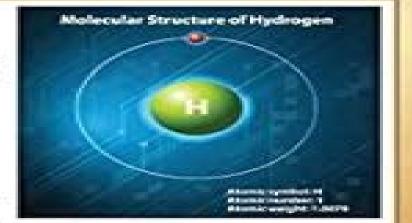
HOW TO BALANCE CHEMPICAL EQUATIONS

A chemical equation is a theoretical or written representation of what hippens during a chemical reaction. The law of conservation of muon states that no atoms can be created at destroyed to a chemical reaction, so the number of atoms that are present in the creatants best to believe the number of atoms that are present in the products. Follow the guide to learn how to believe the experience of atoms that are present to the products.

Marchand, C. of D. Traditional bullance.

- Write down your given equation.
 - -CTM + 62 -+ 9720 + 002
- This traction occurs when propage (CIRR) is burned in the presence of occurs to produce water and carbon discode.
- Write down the number of atoms that you have on each side of the equation.
- Levels at the endouviges result to much alone to find the reunitor of alones in the expension.
 - -Left safe: I carbon, If hydrogen and 2 saygen.
 - -Right side: I corbon, I had regen and I orosen.
- Always leave Indropes and exspect for last. This means that you will need to balance the carbon status first.
- Add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation.

 - The coefficient 3 in front of carbon on the right side indicates 3 carbon stoms just as the subscript 3 as the left side indicates 3 carbon stoms.
 - In a chemical equation, you can change coefficients, but you should never after the subscripts.
- Balance the bedrogen atoms meet. You have 8 on the left side. So you'll soud if on the right side.
 - 400 HER + 100 10 HERO + 10000
 - -On the right side, you new added a 4 as the coefficient because the subscript showed that you already had 2 hydrogen atoms.
 - -When you multiply the coefficient 4 times by the pulsorian 2, you end up with 8.
- The other 6 atoms of Occupes come from 360 subscript 2.03(2+6 atoms of environ- the other 4+00).
- Balance the incigen attents.
 - Because you've added coefficients to the molecules on the right side of the equation, the number of suggest atoms has changed. You new have 6 oxyges atoms in the water molecule and 6 oxyges atoms in the carbon dioxide molecule. That makes a total of 19 oxyges atoms.
 - Whill a coefficient of 5 to the copyes molecule on the left side of the equation. You now have 10 copyes molecules on each state.
 - CHEE + SOC -- SERVE + DOOR.



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