

Chemistry 111 Study Guide Chapters 46-48

Chapter 46

Key Concepts:

- In order to understand the behavior of atoms and molecules, we need to understand the nature of the forces that hold them together.*
- The forces that hold atoms and molecules together are called intermolecular forces.*
- Intermolecular forces can be divided into two categories: van der Waals forces and dipole-dipole interactions.*
- Van der Waals forces are weak forces that arise due to the temporary asymmetries in the distribution of electrons around an atom or molecule.*
- Dipole-dipole interactions are forces that arise between molecules that have permanent dipoles.*

Important Events:

- In 1873, Dutch physicist Johannes Diderik van der Waals first proposed the existence of intermolecular forces.*
- In 1937, American chemist Linus Pauling proposed the dipole-dipole interaction as a type of intermolecular force.*

Notable Figures:

- Johannes Diderik van der Waals: Dutch physicist who first proposed the existence of intermolecular forces.*
- Linus Pauling: American chemist who proposed the dipole-dipole interaction as a type of intermolecular force.*

Chemistry 111 Study Guide

Richard Bailey



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