



ANSWER KEY

How many atoms of **oxygen (O)** are in the **chemical formula** below?

- A 4
B 5
C 7
D 12



(d)

The diagram shown below is a **structural formula** for the gas **propane**. What would be the correct **chemical formula** for propane?

- A C_3H_8
B CH_3
C $8\text{H}_3\text{C}$
D CH



(a)

Carbon (C) and hydrogen (H) create **covalent bonds** to share **electrons** between them. What is the total number of electrons being shared between **carbon and hydrogen** in the molecule of propane shown below?

- A 8 C 16
B 11 D 32



(c)

The **reason** why the bonds between carbon (C) and hydrogen (H) are **covalent** is _____

- A C and H do not have electrons to give away
B C and H are both nonmetals
C C and H are both metalloids
D C and H can bond only by sharing electrons



(d)

The diagram below represents the **structural formula** for several **water molecules**. How many **molecules** are represented?

- A 3
B 5
C 10
D 15



(b)

Explain the **arrangement** of the molecules in the diagram of water below.

- A It is a lucky arrangement
B water molecules are sticky
C water molecules have polarity
D water molecules are magnetic



(c)

Water molecules have some **polarity** in their structure. Using the diagram below, notice how the ends of H atoms (**positive**) and O atoms (**negative**) meet. This property is the **reason** that _____

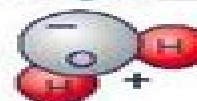
- A water forms droplets easily
B water boils easily
C ice melts easily
D water has color



(a)

Since H atoms are slightly **positive** and the O atoms are slightly **negative**, molecules of water have **polarity**. This makes water a **good** _____

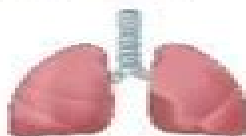
- A solute
B refrigerant
C solvent
D magnet



(c)

The chemical formula for the **oxygen** that we breathe is _____

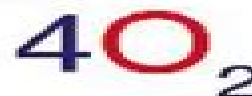
- A O
B O_2
C O_3
D 2O



(b)

How many **molecules of oxygen** are represented in the chemical formula below?

- A 2
B 4
C 6
D 18



(b)

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L Reisser



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