

Sample Questions for the Chemistry Placement Test

The chemistry placement test is used to assess your present level of general chemistry knowledge in addition to your mathematical skills. You will be provided scratch paper (you cannot write on the test itself) and the Periodic Table of the Elements, and you will be allowed to use a calculator. You will have 45 minutes to answer 44 multiple choice questions, with the following sample questions intended only as a guideline:

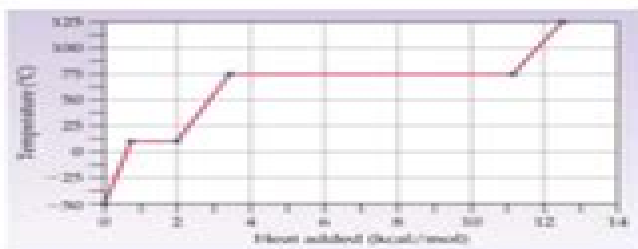
- If the formula for potassium chlorate is KClO_3 and the formula for magnesium fluoride is MgF_2 , then what is the formula for magnesium chlorate?
 - MgClO_3
 - Mg_2ClO_3
 - $\text{Mg}(\text{ClO}_3)_2$
 - $\text{Mg}_2(\text{ClO}_3)_6$
- From the periodic table, what is the atomic number of aluminum?
 - 26.98
 - 13
 - 18
 - 39.95
- Which one of the following elements does not exist as a diatomic molecule in nature?
 - hydrogen
 - nitrogen
 - fluorine
 - neon

For questions 4. and 5., consider the following reaction:



- The reaction can be classified as which one of the following types?
 - precipitation
 - decomposition
 - synthesis
 - double displacement
- How many moles of Al_2O_3 can be produced from the reaction of 10.0 g of Al and 19.0 g of O_2 ?
 - 0.581 mol
 - 0.371 mol
 - 0.185 mol
 - 0.396 mol
- What volume of 12.0 M HCl is required to make 75.0 mL of 3.50 M HCl?
 - 21.9 mL
 - 0.560 mL
 - 257 mL
 - 75.0 mL
- A fish tank holds 1.029 yd^3 of water. What is this volume in cubic meters given that $1 \text{ m} = 1.093 \text{ yd}$?
 - 1.062 m^3
 - 0.9414 m^3
 - 1.125 m^3
 - 0.7881 m^3
- Which one of the following is a strong acid?
 - HNO_3
 - CaSO_4
 - NH_3
 - NaOH

For questions 9. and 10., consider the following heating curve of a hypothetical substance:



- What is the boiling point of the substance?
 - 0°C
 - -50°C
 - 12°C
 - 75°C
- What state of matter is the substance at 50°C ?
 - gas
 - liquid
 - solid
 - not enough information

Answers

1. (c) 2. (b) 3. (d) 4. (c) 5. (c) 6. (a) 7. (d) 8. (a) 9. (d) 10. (b)

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Table of Contents Chemistry Placement Test Bucks Sample

1. Understanding the eBook Chemistry Placement Test Bucks Sample
 - The Rise of Digital Reading Chemistry Placement Test Bucks Sample
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Placement Test Bucks Sample
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Placement Test Bucks Sample
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Placement Test Bucks Sample

- Personalized Recommendations
- Chemistry Placement Test Bucks Sample User Reviews and Ratings
- Chemistry Placement Test Bucks Sample and Bestseller Lists
- 5. Accessing Chemistry Placement Test Bucks Sample Free and Paid eBooks
 - Chemistry Placement Test Bucks Sample Public Domain eBooks
 - Chemistry Placement Test Bucks Sample eBook Subscription Services
 - Chemistry Placement Test Bucks Sample Budget-Friendly Options
- 6. Navigating Chemistry Placement Test Bucks Sample eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Placement Test Bucks Sample Compatibility with Devices
 - Chemistry Placement Test Bucks Sample Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Placement Test Bucks Sample
 - Highlighting and Note-Taking Chemistry Placement Test Bucks Sample
 - Interactive Elements Chemistry Placement Test Bucks Sample
- 8. Staying Engaged with Chemistry Placement Test Bucks Sample
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Placement Test Bucks Sample
- 9. Balancing eBooks and Physical Books Chemistry Placement Test Bucks Sample
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Placement Test Bucks Sample
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Placement Test Bucks Sample
 - Setting Reading Goals Chemistry Placement Test Bucks Sample
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Placement Test Bucks Sample

- Fact-Checking eBook Content of Chemistry Placement Test Bucks Sample
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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