

6.5 Multiple Choice Questions on Spectrometric Methods of Analysis

1. The Beer-Lambert law states that $A = \epsilon cl$. The units of ϵ are
 - (a) cm mol L^{-1}
 - (b) $\text{L mol}^{-1} \text{cm}^{-1}$
 - (c) none – the quantity is dimensionless
 - (d) none of the above
2. If the spectrum of a compound is recorded at different concentrations, which of the following values will change?
 - (a) Absorbance
 - (b) λ_{max}
 - (c) ϵ
 - (d) Absorbance and the λ_{max}
3. The wavenumber of a transition is 2000 cm^{-1} . To what part of the EM spectrum does this belong?
 - (a) IR
 - (b) Microwave
 - (c) Radio wave
 - (d) UV-vis
4. What is the absorbance of a 0.1 mol dm^{-3} aqueous solution of a nickel(II) salt which has $\epsilon = 2.95 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$, and the path length of the solution cell used for the measurement is 1 cm ?
 - (a) 0.295
 - (b) 0.340
 - (c) 29.5
 - (d) 0.029
5. A shift to a lower wavenumber for an absorption peak in a spectrum corresponds to
 - (a) a shift to a higher energy
 - (b) a shift to a lower frequency
 - (c) a shift to a higher frequency
 - (d) a shift to a lower wavelength
6. Which of the following bonds will give the strongest absorption in the IR region?
 - (a) S-H
 - (b) N-H
 - (c) H-H
 - (d) O-H

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AW Rasmussen



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