

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation = $D.DD \times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Midterm Study Guide Answers 2015

Houk

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Chemistry Midterm Study Guide Answers 2015:

The Resistant Learner Lawrence J. Greene, 2015-12-22 Addressing the emotional and academic needs of kids in grades four through twelve Lawrence Greene examines the counterproductive behaviors and attitudes that undermine academic performance diminish self esteem and limit educational and career options Each chapter focuses on a particular deficit area and offers accessible hands on methods for parents to guide and mentor their struggling child Parents learn how to Help their child identify natural talents and use them when learning Choose a democratic or autocratic approach to modifying behavior Use the DIBS system to solve problems Open an effective dialogue with their child Express positive expectations and affirm and praise effort and progress And much more **Surviving Chemistry Guided Study Book** Effiong

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McGraw-Hill's 500 College Chemistry Questions David E. Goldberg, 2012-09-18 500 Ways to Achieve Your Best Grades We want you to succeed on your college chemistry midterm and final exams That s why we ve selected these 500 questions to help you study more effectively use your preparation time wisely and get your best grades These questions are similar to the ones you ll find on a typical college exam so you will know what to expect on test day Each question includes comprehensive explanations in the answer key Whether you have been studying all year or are doing a last minute review McGraw Hill s 500 College Chemistry Questions will help you achieve the final grade you desire Sharpen your subject knowledge and build your test taking confidence with 500 essential college chemistry questions with answers Clear solutions in the answer key for every problem Coverage from atomic mass to electrochemistry **Chemistry 440 - Sample Midterm Exam** Peter

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Kimble, 1994 **Student Study Guide for Chemistry** John E. McMurry, Robert C. Fay, Jill Kirsten Robinson, Julie Klare, Donnajean Fredeen, James Zubricky, 2015-03-09 The Study Guide includes learning goals an overview a review section

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