Exercises

Following the rule you established in the last key question, write correct chemical formula for each of the following compounds

Compound	Formula
(a) Lithium chloride	L." /ce" -> L: Ce
(b) Magnesium iodide	Mg = / I - > Mg Iz
(c) Strontium selenide	51'2/Se2 >> SrSe
(d) Rubidium fluoride	Rb"/F" - RbF
(e) Lithium oxide	L:+1/0-2 -> L:20
(f) Sodium sulfide	Na" 15-2 -> Na, 5
(g) Potassium chloride	K+1/ce-1 -> Kce
(h) Calcium phosphide	Ca+2/P-3 → Ca2P2
(i) Barium oxide	Bc2/0-2 > Ba0
(j) Aluminum sulfide	AQ+3/5-2 -> AD252

3. Use your answers to the Key Questions and the Exercise Questions to draw a conclusion about the ratio of ions in two compounds if the elements in the compounds are from the same groups (example: aluminum oxide and aluminum sulfide; lithium chloride and potassium chloride).

Chemistry A Bonding Packet Answers

Gerhard Quinkert

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