

## HONORS CHEM 1st SEMESTER EXAM STUDY GUIDE FALL 2010

**STRATEGY:** Start by reading through your notes to refresh your memory on these topics. Then, use this review sheet as a starting point to identify the areas on which you need to spend more study time. For those areas, go back to homework assignments, quizzes, and reviews to practice more problems. I would also recommend going through all of your tests since these questions are only samples and do not include specific examples of how vocabulary and other conceptual information might appear in a multiple-choice or other format. Remember you can access notes and reviews on my website.

**IMPORTANT INFORMATION:**

- This must be handed in BEFORE taking your final exam.
- If entirely complete, 5 points will be added to your exam grade
- All work must be shown in DETAIL on lined paper

**Introduction to Chemistry – Chapter 1**

1. Identify safe/unsafe lab practices.
2. List pieces and uses of lab equipment.
3. List the requirements for tables and graphs.
4. What is plotted on the x-axis and on the y-axis?
5. Make a density graph.
6. What is water displacement?

**Matter and Change – Chapter 2**

7. Classify the following substances as *solid*, *liquid*, *gas*, or *plasma* based on their properties.
  - a. flexible volume, high KE, particles can disperse freely.
  - b. flexible volume, very high KE, particles are charged.
  - c. fixed volume, very low KE, orderly particles.
  - d. fixed volume, low KE, particles can move past each other.
8. Classify the following as element, compound, heterogeneous mixture, or solution.
  - a. graphite (carbon)
  - b. grape juice
  - c. table salt (NaCl)
  - d. pepper
9. Classify the following as *chemical* or *physical changes*.
  - a. cutting wire
  - b. ripening tomato
  - c. apple slices turning brown
  - d. compressing a gas
10. Classify the following properties as *physical* or *chemical*.
  - a. melts at  $68.0^{\circ}\text{C}$
  - b. corrosive
  - c. reacts violently with water
  - d. decomposes in air
  - e. magnetic
11. Classify the following properties as *intensive* or *extensive*.
  - a. melting point
  - b. volume
  - c. mass
  - d. reactivity
  - e. density

**VOCAB:** kinetic molecular theory

# Chemistry 1st Semester Exam Study Guide

**Dr. Abhishek M**



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## **Table of Contents Chemistry 1st Semester Exam Study Guide**

1. Understanding the eBook Chemistry 1st Semester Exam Study Guide
  - The Rise of Digital Reading Chemistry 1st Semester Exam Study Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1st Semester Exam Study Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry 1st Semester Exam Study Guide
  - User-Friendly Interface

4. Exploring eBook Recommendations from Chemistry 1st Semester Exam Study Guide
  - Personalized Recommendations
  - Chemistry 1st Semester Exam Study Guide User Reviews and Ratings
  - Chemistry 1st Semester Exam Study Guide and Bestseller Lists
5. Accessing Chemistry 1st Semester Exam Study Guide Free and Paid eBooks
  - Chemistry 1st Semester Exam Study Guide Public Domain eBooks
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  - Chemistry 1st Semester Exam Study Guide Budget-Friendly Options
6. Navigating Chemistry 1st Semester Exam Study Guide eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry 1st Semester Exam Study Guide Compatibility with Devices
  - Chemistry 1st Semester Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry 1st Semester Exam Study Guide
  - Highlighting and Note-Taking Chemistry 1st Semester Exam Study Guide
  - Interactive Elements Chemistry 1st Semester Exam Study Guide
8. Staying Engaged with Chemistry 1st Semester Exam Study Guide
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry 1st Semester Exam Study Guide
9. Balancing eBooks and Physical Books Chemistry 1st Semester Exam Study Guide
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry 1st Semester Exam Study Guide
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 1st Semester Exam Study Guide
  - Setting Reading Goals Chemistry 1st Semester Exam Study Guide
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry 1st Semester Exam Study Guide
  - Fact-Checking eBook Content of Chemistry 1st Semester Exam Study Guide
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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