

1. This question refers to the elements in the first four periods of the Periodic Table.

																H											He
Li	Be																	B	C	N	O	F	Ne				
Na	Mg																	Al	Si	P	S	Cl	Ar				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr										

Identify an element from the first four periods that fits each of the following descriptions.

- (i) The element that forms a $2-$ ion with the same electronic configuration as Ar.

.....

[1]

- (ii) The element that forms a $3+$ ion with ten electrons.

.....

[1]

- (iii) An element that forms a compound with fluorine with trigonal planar molecules.

.....

[1]

- (iv) The element that forms a chloride XCl_2 with a molar mass of 111.1 g mol^{-1} .

.....

[1]

- (v) The element with the largest atomic radius.

.....

[1]

- (vi) The element with the smallest first ionisation energy.

.....

[1]

2. This question refers to the elements in the first four periods of the Periodic Table.

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