

The Nuclear Atom

(What Is an Atom?)

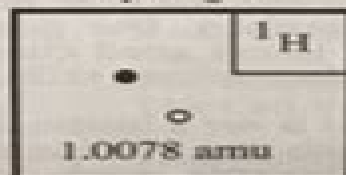
Model: Schematic Diagrams for Various Atoms and Ions.

- electron (-)
- proton (+)
- ◉ neutron (no charge)

$$1 \text{ amu} = 1.6606 \times 10^{-24} \text{ g}$$

The **nucleus** of an atom contains the protons and the neutrons.

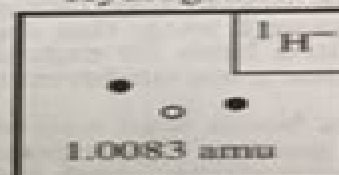
Hydrogen



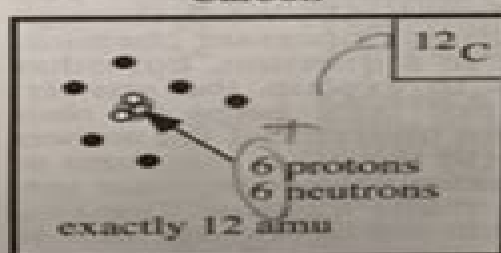
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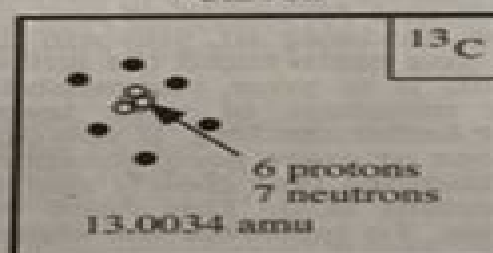
Hydrogen ion



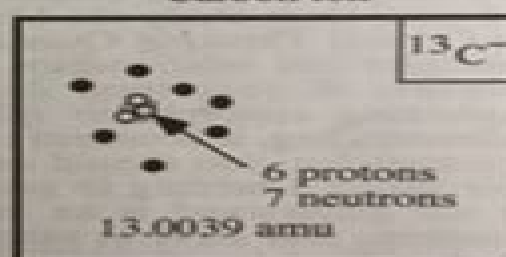
Carbon



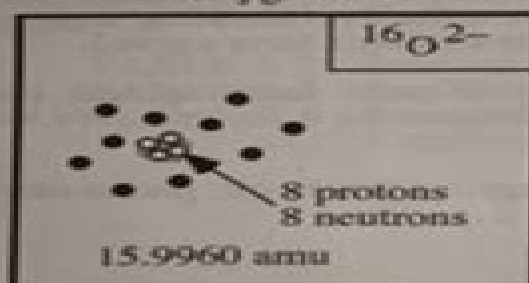
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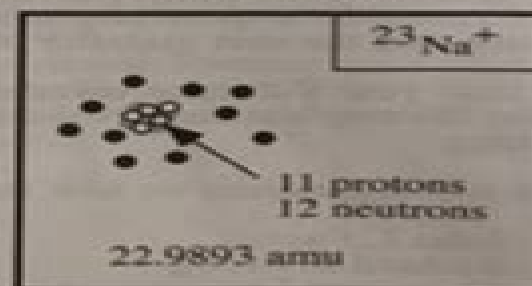
Carbon ion



Oxygen ion



Sodium ion



^1H and ^2H are **isotopes** of hydrogen.

^{12}C and ^{13}C are **isotopes** of carbon.

Chemactivity 6 Atomic Size Answers

Clemens Wendtner



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