

PRACTICE EXAM 2

CHEM 1411 (Chap. 1, 2, 3, 4, 5)

- d 1. Which of the following is a physical change?
 - (a) rusting a nail
 - (b) fruits ripen
 - (c) baking a cake
 - (d) ice melts
 - (e) milk turns sour
- e 2. Which of the following is a homogeneous mixture?
 - (a) water and sand
 - (b) sand, alcohol, and sugar
 - (c) oil and vinegar
 - (d) milk of magnesium
 - (e) air
- e 3. Which of the following is a correct?
 - (a) sodium, S
 - (b) mercury, Mg
 - (c) tin, N, Sn
 - (d) potassium, P
 - (e) gold, Au
- b 4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?
 - (a) 41
 - (b) 314
 - (c) 378
 - (d) 457
 - (e) -32
- a 5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?

$d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$

 - (a) 1.59
 - (b) 1.74
 - (c) 2.36
 - (d) 2.89
 - (e) 3.50
- e 6. How many significant figures are there in the measurement of 1.0356×10^4 ?
 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- c 7. How many significant figures are appropriate to show in the result after carrying out the operation below?

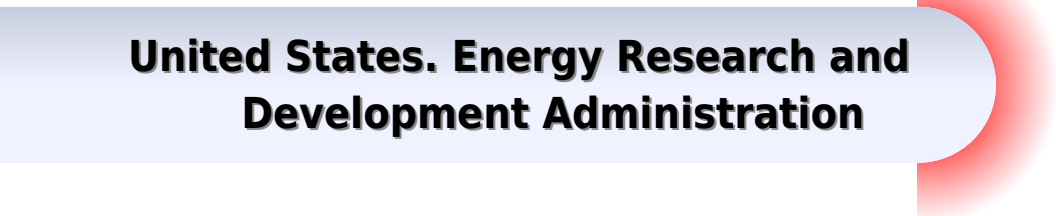
$(13.7 + 0.027) + 8.221 = ?$ $\times, \div$ least s.f. $+, -$ least decimal

13.727

 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- d 8. How many protons, electrons, and neutrons are there in $^{130}\text{Ba}^{2+}$?
 - (a) 56p, 56e, 130n
 - (b) 56p, 54e, 130n
 - (c) 56p, 58e, 74n
 - (d) 56p, 54e, 74n
 - (e) 54p, 56e, 74n
- c 9. Which of the following pairs would have similar chemical and physical properties?
 - (a) Na, Mg
 - (b) H, Li
 - (c) He, Ne
 - (d) Cr, Ca
 - (e) Cl, O
- c 10. Which of the following is the molecular formula?
 - (a) NaCl
 - (b) CsF
 - (c) HCl
 - (d) AlBr_3
 - (e) NaCN
- e 11. Which of the following is the empirical formula?
 - (a) C_2H_4
 - (b) CH_3COOH
 - (c) P_2O_{10}
 - (d) NH_4NO_3
 - (e) Ag_2CO_3
- e 12. Which of the following is a correct match?
 - (a) BaCl_2 , barium dichloride
 - (b) CS_2 , carbon sulfide
 - (c) CuF_2 , copper (I) fluoride
 - (d) HNO_2 , nitrous acid
 - (e) $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, magnesium sulfate heptahydrate
- b 13. Which of the following pairs is a correct match?
 - (a) $\text{Ca}_3(\text{PO}_4)_2$, calcium (II) phosphate
 - (c) Mg_2O_3 , magnesium oxide
 - (b) $(\text{NH}_4)_2\text{SO}_4$, ammonium sulfate
 - (e) H_2CO_3 , carbon acid
 - (d) $\text{Cu}(\text{NO}_3)_2$, copper dinitrate

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