

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A solution of the perfect... 2. A solution of... 3. A solution of... 4. The gas is at... 5. The volume...	Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A solution of the perfect... 2. A solution of... 3. A solution of... 4. The gas is at... 5. The volume...

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Name: _____

Unit 1: Measurement Analysis

Mark the best answer from those provided to each scenario.

Mark the best answer from those provided to each scenario.

Determine the number of Significant Figures in each of the following numbers:

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.000 _____ L	Volume = 1.0000 _____ L

Convert the following into Scientific Notation:

Mass = 1. _____ kg	Mass = 1.00 _____ kg
Volume = 1.000 _____ L	Volume = 1.0000 _____ L

Measurement Analysis Round your answers to 3 places after the decimal. Show work to show your work.

a. Convert the 1000 meters into feet.

b. Convert 1000 meters into miles.

Unit 2: Reaction

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

H_2SO_4

Atomic Masses:

H=1.0
S=32.1
O=16.0

Reaction

Atomic Masses:

H=1.0
S=32.1
O=16.0

Unit 3: Stoichiometry

Write the chemical formula for each compound.

Compound	Formula
Hydrogen	
Carbon	
Oxygen	
Nitrogen	
Sulfur	

Write the chemical formula for each compound.

Compound	Formula
Hydrogen	
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Oxygen	
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Sulfur	

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Chemistry Spring Final Study Guide

Richard Bailey



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