Solution Concentration

Study Guide

Key Concepts

- Concentrations can be measured qualitatively and quantitatively.
- Molarity is the number of moles of solute dissolved per liter of solution.

molarity
$$(M) = \frac{\text{moles of solute}}{\text{liters of solution}}$$

 Molality is the ratio of the number of moles of solute dissolved in 1 kg of solvent.

molality
$$(m) = \frac{\text{moles of solute}}{\text{kg of solvent}}$$







Chemistry Solution Concentration Study Guide

David R. Klein, Laurie S. Starkey

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Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition John McMurry, Susan McMurry, 1999

Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

Study Guide and Solutions Manual Neil E. Schore, 2002-08-02

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