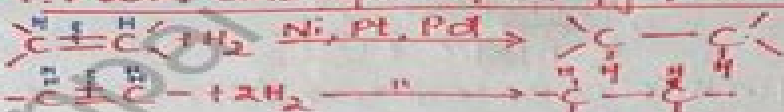


HYDROCARBON

MOP of Alkane

1. From unsaturated hydrocarbon



DM $\rightarrow$ π bond ko H.H se tod do.

exothermic

4 MCTS formed

Syn addition

Addⁿ of conj. diene



$\text{RaNi} \rightarrow$ alloy of Ni & Al in NaOH
Galvanom temp better than Ni (at 100°C)

If RaNi used $\rightarrow$ Sabatier Senderason's

$[\text{Rh}(\text{PPh}_3)_3\text{Cl}]$ used also

(Wilkinson's catalyst)

Alkene	Addition	Product
Cis	Syn	Erythro
Trans	Syn	Threo
Cis	Anti	Threo
Trans	Anti	Erythro

2. From alkyl halide



X hatoo H lagao.

$\text{ROR} : \text{R} - \text{Cl} < \text{R} - \text{Br} < \text{R} - \text{I}$ $\text{ROR} \propto \frac{1}{\text{BE}}$

3. Wurtz reaction



produce higher alkane with even no. of 'C'

$\text{R} - \text{X} \rightarrow$ coupling

$\text{R} - \text{X} \rightarrow$ disproportionation

in 2 molecular wurtz faster than in 1 molecular but prod should be stable i.e A.A nahi banana hai.

$\text{ROR} \propto \frac{1}{\text{SH}}$ $\text{R} - \text{X} > 2^\circ \text{R} - \text{X} > 3^\circ \text{R} - \text{X}$

for same R: $\text{RF} < \text{RCI} < \text{RBr} < \text{RI}$

4. Decarboxylation reaction



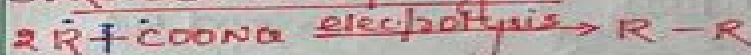
- COOH hatoo H lagao



- COONa hatoo H lagao

$\text{ROR} \propto$ stability of intermediate (R^-)

5. Kolbe's electrolysis



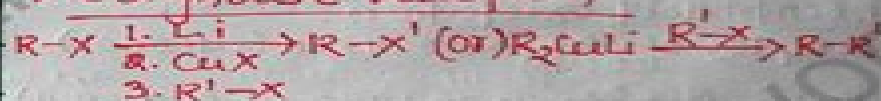
Cathode $\rightarrow \text{H}_2$ Anode $\rightarrow$ Alkane & CO_2

Addition Reaction

6. Frankland reaction



7. Coreyhouse reaction



Attack of $\text{R}^- \rightarrow \text{RDS}$

$\text{ROR} \propto \frac{1}{\text{SH of R}' - \text{X}}$

If same R:

$\text{R} - \text{F} < \text{R} - \text{Cl} < \text{R} - \text{Br} < \text{R} - \text{I}$

$\text{BE} \downarrow \rightarrow$ leaving tendency $\uparrow$

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Neil E. Schore



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1. Understanding the eBook Chemistry Study Guide Answer Key Hydrocarbons
 - The Rise of Digital Reading Chemistry Study Guide Answer Key Hydrocarbons
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Answer Key Hydrocarbons
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Study Guide Answer Key Hydrocarbons
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Answer Key Hydrocarbons
 - Personalized Recommendations
 - Chemistry Study Guide Answer Key Hydrocarbons User Reviews and Ratings
 - Chemistry Study Guide Answer Key Hydrocarbons and Bestseller Lists

5. Accessing Chemistry Study Guide Answer Key Hydrocarbons Free and Paid eBooks
 - Chemistry Study Guide Answer Key Hydrocarbons Public Domain eBooks
 - Chemistry Study Guide Answer Key Hydrocarbons eBook Subscription Services
 - Chemistry Study Guide Answer Key Hydrocarbons Budget-Friendly Options
6. Navigating Chemistry Study Guide Answer Key Hydrocarbons eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Study Guide Answer Key Hydrocarbons Compatibility with Devices
 - Chemistry Study Guide Answer Key Hydrocarbons Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Study Guide Answer Key Hydrocarbons
 - Highlighting and Note-Taking Chemistry Study Guide Answer Key Hydrocarbons
 - Interactive Elements Chemistry Study Guide Answer Key Hydrocarbons
8. Staying Engaged with Chemistry Study Guide Answer Key Hydrocarbons
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Study Guide Answer Key Hydrocarbons
9. Balancing eBooks and Physical Books Chemistry Study Guide Answer Key Hydrocarbons
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Study Guide Answer Key Hydrocarbons
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Study Guide Answer Key Hydrocarbons
 - Setting Reading Goals Chemistry Study Guide Answer Key Hydrocarbons
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Study Guide Answer Key Hydrocarbons
 - Fact-Checking eBook Content of Chemistry Study Guide Answer Key Hydrocarbons
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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