

SOLUTION

1.	Solution: A homogeneous mixture of two
	or more substances is known as solution.
2.	Solute: The substance present in smaller amount in a solution is called solute.
3.	Solvent: The substance present in larger

- Solvent: The substance present in larger amount in a solution is called solvent.
- Binary solution: Theose solutions which contain two components are called binary solutions, e.g., salt solution, benzene and toluene.
- Ternary solutions: Those solutions which contain three components are called ternary solutions, e.g., ethanol + water + acetic acid.
- Aqueous solution: When solute is dissolved in water, it is called aqueous solution, e.g., sugar solution, ethanol in water.
- Non-aqueous solution: When solute is dissolved in solvent other than water, it is called non-aqueous solution, e.g., iodine dissolved in alcohol (Tincture of iodine).
- 8. Weight Fraction = weight of solution
- Weight percentage

Weight of solute present in 100 grams of solution is called weight percentage

Volume of solute present in 100 ml of the solution is called volume percentage

- 11. Volume Fraction = Volume of solution
- 12. Molarity (M)

(or)
$$M = \frac{W}{GMW} \times \frac{1}{V(lit)}$$

13. Molarity = No of moles of solute

14. Molarity = $\frac{\text{Weight of solute}}{\text{GMWt of solute}} \times \frac{1000}{\text{Vin ml}}$

(or)
$$M = \frac{W}{GMW} \times \frac{1000}{V(ml)}$$

- 15. Weight of solute = $\frac{M \times GMW \times Vinml}{1000}$
- 16. Molarity for dilute solutions: $M_iV_i = M_iV_j$
- 17. For neutralisation reactions: $\frac{M_1V_1}{n_1} = \frac{M_2V_2}{n_2}$
- 18. For mixed solution molarity

$$= \frac{M_1V_1 + M_2V_2 +}{V_1 + V_2 +}$$

Chemistry Dimensions 2 Solutions

Peter Atkins, Julio de Paula, Ron Friedman

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Career Point Kota JEE Main 2020 Chapterwise Solved Papers Physics, Chemistry, and Mathematics Career Point Kota, 2022-03-27 Here s introducing the all new edition of 2020 JEE Main Chapterwise Solved Papers this book has been comprehensively comprised of all 16 Sets of online papers that were conducted in January September 2020 Giving complete detailed and authentic solutions to all the questions this book serves as a must have practice manual before the final call in the examination hall Whenever a student decides to prepare for any examination her his first and foremost curiosity about the type of questions that he she has to face This becomes more important in the context of competitive examinations where there is neck to neck race We feel great pleasure to present before you this book We have made an attempt to provide chapter wise questions asked in JEE Main 2020 all 16 sets of January September attempts with solutions Solutions to the questions are not just sketch rather have been written in such a manner that the students will be able to under the application of concept and can answer some other related questions too We firmly believe that the book in this form will definitely help a genuine hardworking student We have tried our best to keep errors out of this book Comment and criticism from readers will be highly appreciated and incorporated in the subsequent edition We wish to utilize the opportunity to place on record our special thanks to all team members of Content Development for their efforts to make this wonderful book

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Marie Claire Bellissent Funel and George Neilson I am sure that all the participants would wish to congratulate our two colleagues for putting together an outstanding programme of lectures round tables and poster session The theory which underlies the subject was covered by four leading authorities J P Hansen Paris set out the general framework in terms of the statistical mechanics of bulk and surface properties H L Friedman Stony Brook focused attention on ionic liquids at equilibrium and I B Hubbard considered non equilibrium properties such as the electrical conductivity and ionic friction coefficients Finally the basic theory of polyelectrolytes treated as charged linear polymers in aqueous solution was presented by I M Victor Paris Physical Chemistry for Students of Medicine Alexander Findlay, 1924 The Fractal Physical Chemistry of Polymer Solutions and Melts G. V. Kozlov, I. V. Doblin, Gennady E. Zaikov, 2013-12-12 This book provides an important structural analysis of polymer solutions and melts using fractal analysis. The book covers the theoretical fundamentals of macromolecules fractal analysis It then goes on to discuss the fractal physics of polymer solutions and the fractal physics of melts The intended audience of the book includes specialists in chemistry and physics of polymer synthesis and those in the field of polymers and polymer composites processing Geopolymer, Green Chemistry and Sustainable **Development Solutions** Joseph Davidovits, 2005 **Solutions Manual for Quanta, Matter and Change Peter** Atkins, Julio de Paula, Ron Friedman, 2008-12-15 Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

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