Chemistry Cookie Project - Chocolate Chip

- In this lab you will be converting a recipe from moles to standard cooking measurements and then using that recipe to bake some cookies!
- You will need the following tables in order to convert your recipe:

Molecular Formula Table

Use the following molecular formulas to calculate grams of ingredients.

NOTE: Most of these substances have extremely complex molecular formulas.

I have greatly simplified your project by listing a representative formula only.

Anise	C ₁₀ H ₁₂ O
Baking Soda	NaHCO ₃
Baking Powder	NaHCO ₃
Brown Sugar	$C_{12}H_{22}O_{11}$
Butter	C ₉ H ₁₄ O ₆
White (Cane) Sugar	C12H22O11
Chocolate	C ₄ H ₈ O ₄
Cinnamon	C ₉ H ₈ O
Cloves	C ₁₀ H ₁₂ O ₂
Cream of tartar	KHC ₄ H ₅ O ₆

Flour	C ₄ H ₈ O ₄
Lemon Juice (extract)	C ₆ H ₈ O ₇
Margarine	C9H12O6
Milk	85% H ₂ O 15% C ₀ H ₁₄ O ₆
Molasses	$C_{12}H_{22}O_{11}$
Pepper	C ₁₇ H ₁₉ O ₃ N
Salt	NaCl
Vanilla	C ₈ H ₈ O ₃
Eggs	C ₆ H ₁₂ O ₃ N ₂

Unit Conversions

Use the following conversions to get from grams to a standard cooking unit of measure.

1 teaspoon of baking soda = 2.84g 1 teaspoon of vanilla extract = 4.73g 1 large egg = 50g 1 cup flour = 141.95 g 1 teaspoon salt= 4.16 g 1 cup butter = 236.59 g 1 cup sugar = 198.73 g 1 cup brown sugar = 141.46 g 1 ounce chocolate chips = 28.35 g 1 tablespoon lemon juice = 14.20 g 1 teaspoon baking powder = 2.84 g 1 teaspoon cream of tartar = 2.84 g 1 teaspoon cinnamon = 2.84 g

<u>Chemistry Cookie Project Chocolate Chip Answer Key</u> <u>Common Core</u>

A Gutmann

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