

CHEMISTRY WORKSHEET # 2: THE MOLE AS A UNIT OF MASS

Now that you know how to find the mass of one mole of a substance (molar mass) you can easily find the mass of several moles or the mass of a fraction of a mole using the factor-label technique.

1 mol = a molar mass of an atom/molecule (g/mol)

EXAMPLE: What is the mass of 5.00 moles of water H_2O ?

$$\text{STEP 1: } 2 \text{ H} = 2 \times (1.01) = 2.02$$

$$\text{O} = 1 \times (16.00) = 16.00$$

$$H_2O = 18.02 \text{ g}$$

$$\text{STEP 2: } 5.00 \text{ moles } H_2O \times \frac{18.02 \text{ g } H_2O}{1 \text{ mole } H_2O} = 90.10 \text{ g } H_2O$$

NOW YOU TRY ONE: What is the mass of 0.50 moles of calcium carbonate $CaCO_3$?

$$\text{STEP 1: } Ca =$$

$$C =$$

$$3O =$$

$$CaCO_3 =$$

$$\text{STEP 2: } 0.50 \text{ g } CaCO_3 =$$

USE A SEPARATE SHEET OF PAPER TO SOLVE THE FOLLOWING PROBLEMS. SHOW YOUR WORK. ROUND MOLAR MASSES TO TWO PLACES AFTER THE DECIMAL. ADD UNITS.

1. How many grams are there in 5.00 moles of lead Pb ?

$$310.8 \text{ g } Pb$$

2. How many grams are there in 2.00 moles of sulfuric acid H_2SO_4 ?

$$196.16 \text{ g } H_2SO_4$$

3. How many grams are there in 0.250 moles of sodium hydroxide $NaOH$?

$$10 \text{ g } NaOH$$

4. How many grams are there in 2.50 moles of potassium nitrate KNO_3 ?

$$252.75 \text{ g } KNO_3$$

5. How many grams are there in 10.0 moles of lithium carbonate Li_2CO_3 ?

$$738.9 \text{ g } Li_2CO_3$$

Now that you know how to find the mass of one mole of a substance you can easily find the number of moles there are in a given mass of the substance.

EXAMPLE: How many moles of calcium chloride are there in 333 grams of calcium chloride ($CaCl_2$)?

$$\text{STEP 1: } Ca = 1 \times (40.08) = 40.0$$

$$2 \text{ Cl} = 2 \times (35.45) = 70.90$$

$$CaCl_2 = 110.98 \text{ g}$$

$$\text{STEP 2: } 333 \text{ grams } CaCl_2 \times \frac{1 \text{ mole } CaCl_2}{110.98 \text{ g } CaCl_2} = 3.00 \text{ mole } CaCl_2$$

USE THE SAME PAPER AS THE ABOVE PROBLEMS TO SOLVE THE FOLLOWING. SHOW YOUR WORK AND PUT UNITS ON EACH ANSWER!

6. How many moles of silver nitrate are there in 80.00 grams of silver nitrate $AgNO_3$?

$$0.47 \text{ mol } AgNO_3$$

7. How many moles of phosphoric acid are there in 658 grams of phosphoric acid H_3PO_4 ?

$$6.71 \text{ mol } H_3PO_4$$

8. How many moles of tin (II) fluoride are there in 908 grams of tin (II) fluoride SnF_2 ?

$$5.79 \text{ mol } SnF_2$$

9. How many moles of hydrogen peroxide are there in 1000.0 grams of hydrogen peroxide H_2O_2 ?

$$29.39 \text{ mol } H_2O_2$$

10. How many moles of magnesium chloride are there in 148 grams of magnesium chloride $MgCl_2$?

$$1.55 \text{ mol } MgCl_2$$

#1-5: You are given the number of moles and asked to find the grams (mass)

MOLS $\rightarrow$ GRAMS

#6-10: You are given the grams (mass) and asked to find the number of mols

GRAMS $\rightarrow$ MOLS

Chemistry B Moles Packet

Karin Nielsen-Saines



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Chemistry B Moles Packet: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the captivating narratives that have enthralled audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Uncover the Best : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens' "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and thrilling novel that will keep you guessing until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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