

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a heterogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution and an unsaturated solution.

Define the term solvent and provide an example.

What is solubility, and how is it measured?

Distinguish between concentrated and dilute.

What are the properties of acids?

What are the properties of bases?

What is the pH scale?

What is the primary characteristic of acids?

What is the primary characteristic of bases?

What is the pH of a neutral solution?

What is the pH of a strong acid?

What are the properties of bases and how are they measured?

Explain the significance of the pH scale.

How does the concentration of an acid or base affect its pH?

What happens to the pH of a solution during neutralization?

What happens when strong acids and bases are mixed?

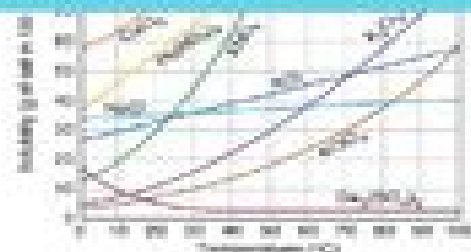
What is the effect of water on the pH of a solution?

What is the effect of temperature on the pH of a solution?

What is the effect of dilution on the pH of a solution?

What is the effect of concentration on the pH of a solution?

What is the effect of the nature of the acid or base on the pH of a solution?



20. Describe how you would use the graph to determine the amount of Na_2SO_4 that can be dissolved in 100 grams of water at 40°C .

22. Which salt has the highest solubility at 40°C ?



ANSWER KEY INCLUDED

Chemistry Acids And Bases Study Guide

David R. Klein, Laurie S. Starkey



Chemistry Acids And Bases Study Guide:

Acid-Base Equilibria - Quick Chemistry Review Outline and Handout E Staff, Acid Base Equilibria Quick Review Outline and Handout for All Students Learn and review on the go Use Quick Review Chemistry Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students 10 Pages

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