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#### Chemistry 1411 Practice Exam 1, Chapters 1-3

Show your calculations clearly on show work problems. A periodic table is also included. Good luck!

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1	China there	mauroubsert of	stignificant.	dignames in th	he following	numbers (3 pts)	äc.

1) 6.022 X 10<sup>23</sup> atoms/mol 2) 0.049001 g

a) 3, 5

b) 4, 2

c) 23, 7

d) 4, 5

A rectangular block with a length, width, and height of 2.55 cm, 2.55 cm, and 1.44 cm has a density of 0.850 g/cm<sup>3</sup>. What is the mass of this object in grams? (3 pts)
 a) 5.46 g
 b) 7.96 g
 c) 8.65 g
 d) 10.4 g

(3 pts) Isotopes are atoms that:

a) have different atomic numbers

b) have the same mass number

c) have the same number of protons but different numbers of neutrons

d) have different numbers of electrons

4. A general summary based on observed and measurable behavior in many different systems

is called a: (3 pts)
a) hypothesis

b) natural law

c) model.

d) theory

What is the number of protons, neutrons, and electrons in the iron-56 atom? (3 pts)

a) 56 protons; 26 neutrons; 24 electrons b) 26 protons; 30 neutrons; 24 electrons

c) 24 protons; 56 neutrons; 26 electrons d) 26 protons, 30 neutrons; 26 electrons

Which of the following pairs of elements is a <u>transition element</u> and a <u>halogen</u>? (3 pts)

a) W and Si

b) Cr and Ar

c) Au and Br

d) Ba and U

Which of the following compounds is molecular? (3 pts)

a) BarCla

b) NaOH

c) SFa

d) KNOs

Which of the following elements is a noble gas? (3 pts)

a) Mg

b) Xe

e) Pb

d) Ber

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Judd E. Hollander

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- 1. Understanding the eBook Chemistry 1411 Lab Answers
  - o The Rise of Digital Reading Chemistry 1411 Lab Answers
  - o Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry 1411 Lab Answers
  - Exploring Different Genres
  - $\circ\,$  Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry 1411 Lab Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry 1411 Lab Answers
  - Personalized Recommendations
  - Chemistry 1411 Lab Answers User Reviews and Ratings
  - Chemistry 1411 Lab Answers and Bestseller Lists
- 5. Accessing Chemistry 1411 Lab Answers Free and Paid eBooks
  - Chemistry 1411 Lab Answers Public Domain eBooks
  - Chemistry 1411 Lab Answers eBook Subscription Services
  - Chemistry 1411 Lab Answers Budget-Friendly Options
- 6. Navigating Chemistry 1411 Lab Answers eBook Formats
  - o ePub, PDF, MOBI, and More
  - Chemistry 1411 Lab Answers Compatibility with Devices
  - Chemistry 1411 Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry 1411 Lab Answers
  - Highlighting and Note-Taking Chemistry 1411 Lab Answers
  - Interactive Elements Chemistry 1411 Lab Answers
- 8. Staying Engaged with Chemistry 1411 Lab Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Chemistry 1411 Lab Answers
- 9. Balancing eBooks and Physical Books Chemistry 1411 Lab Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry 1411 Lab Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 1411 Lab Answers
  - Setting Reading Goals Chemistry 1411 Lab Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 1411 Lab Answers
  - Fact-Checking eBook Content of Chemistry 1411 Lab Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - o Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - $\circ \ \ Integration \ of \ Multimedia \ Elements$
  - Interactive and Gamified eBooks

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