

Exercises

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1. Which of the following reactions are exothermic?

(a) $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

(b) $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$

(c) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

(d) $\text{H}_2\text{O} + \text{CO} \rightarrow \text{H}_2 + \text{CO}_2$

2. Which of the following reactions are endothermic?

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(a) $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

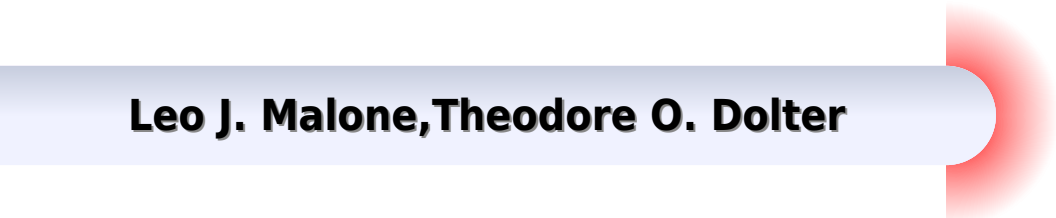
(b) $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$

(c) $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

(d) $\text{H}_2\text{O} + \text{CO} \rightarrow \text{H}_2 + \text{CO}_2$

Chemical Reaction Equation Study Guide With Answers

Leo J. Malone, Theodore O. Dolter



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