

CHEMICAL QUANTITIES STUDY GUIDE

Atomic Mass

- The mass of a single atom of an element (amu)
- Mass of one mole of the atom = atomic mass x amu
- $1 \text{ amu} = 1.66 \times 10^{-24} \text{ g}$

Avogadro's Number

- 6.022×10^{23}
- Six hundred 2 sextillion two hundred quintillion
- Represents the number of atoms that are in 1 mole of a substance

Molar Mass

- The sum of the total mass in grams of all the atoms that make up a mole of a particular molecule
- Molar mass of an element = number underneath name on periodic table
 - For a compound, calculate molar mass of each element and add

Mole Conversions

- Moles = mass/molar mass

$$\begin{array}{ccccc} \text{MASS (g)} & \begin{array}{c} \times \\ \hline \text{MOLAR MASS} \\ \hline + \end{array} & \boxed{\text{MOLE}} & \begin{array}{c} \times \\ \hline 6.022 \times 10^{23} \\ \hline + \end{array} & \text{FORMULA} \\ & & \begin{array}{c} \times \quad | \quad 22.4 \quad | \quad + \\ \hline \text{VOLUME} \end{array} & & \text{UNITS} \end{array}$$

Chemical Quantities Study Guide

Julie R. Frentrup



Chemical Quantities Study Guide:

Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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