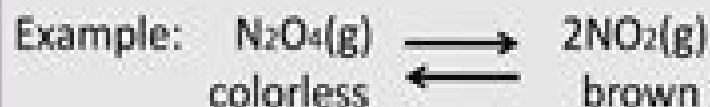


# CHEMICAL EQUILIBRIUM

A state or condition in which two opposing chemical reaction occur simultaneously at the same rate. It is a state where the concentrations of reactants and products remain constant over time.

Chemical equilibrium is achieved when the rates of forward and reverse reactions are equal and the concentrations of the reactants and products remain constant.

Equilibrium mixture – a mixture of reactants and products in the equilibrium state.



# Chemical Equilibrium Study Guide

**John Krenos, Joseph Potenza, Carl  
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Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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