

## Chemistry Bonding MCQ (From 2015 Prelim Papers)

### Q1

In which of the following pairs does the first species have a larger bond angle than the second?

- A  $\text{SO}_2$ ,  $\text{PI}_3$
- B  $\text{SO}_3^{2-}$ ,  $\text{SO}_4^{2-}$
- C  $\text{H}_2\text{S}$ ,  $\text{NH}_3$
- D  $\text{H}_2\text{O}$ ,  $\text{H}_3\text{O}^+$

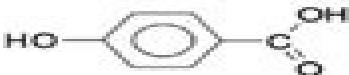
### Q2

Which of the following statements is correct?

- A  $\text{BCl}_3$  and  $\text{BrCl}_3$  have the same molecular shape.
- B Cyclohexene and propanone are non-polar liquids.
- C  $\text{LiAlH}_4$  and  $\text{NaBF}_4$  contain dative covalent bonds.
- D  $\text{XeF}_4$  and  $\text{H}_2\text{C}=\text{C}=\text{CH}_2$  are planar molecules.

### Q3

In which of the following pairs does the first substance have a higher melting point than the second?

|   | First substance                                                                   | Second substance                                                                    |
|---|-----------------------------------------------------------------------------------|-------------------------------------------------------------------------------------|
| A | $\text{CH}_3\text{CH}_2\text{OCH}_3$                                              | $\text{CH}_3\text{CH}_2\text{NHCH}_3$                                               |
| B | $\text{CH}_3(\text{CH}_2)_4\text{CH}_3$                                           | $(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)_2$                                         |
| C | $\text{RbCl}$                                                                     | $\text{KCl}$                                                                        |
| D |  |  |

### Q4

Which of the following is a planar molecule?

- A  $\text{C}_2\text{F}_4^{2+}$
- B  $\text{CH}_2=\text{C}=\text{CH}_2$
- C 
- D  $\text{CH}_2=\text{CH}-\text{CHO}$

# Chemical Bonding Multiple Choice Test

**Arshad Iqbal**



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- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chemical Bonding Multiple Choice Test. The first chapter will explore what Chemical Bonding Multiple Choice Test is, why Chemical Bonding Multiple Choice Test is vital, and how to effectively learn about Chemical Bonding Multiple Choice Test.

3. In chapter 2, this book will delve into the foundational concepts of Chemical Bonding Multiple Choice Test. This chapter will elucidate the essential principles that need to be understood to grasp Chemical Bonding Multiple Choice Test in its entirety.

4. In chapter 3, the author will examine the practical applications of Chemical Bonding Multiple Choice Test in daily life. This chapter will showcase real-world examples of how Chemical Bonding Multiple Choice Test can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Chemical Bonding Multiple Choice Test in specific contexts. The fourth chapter will explore how Chemical Bonding Multiple Choice Test is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Chemical Bonding Multiple Choice Test. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chemical Bonding Multiple Choice Test.

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