3.
$$S_6 + \frac{12}{12} O_2 \rightarrow \frac{8}{12} SO_3$$
4. $\frac{1}{2} N_2 + \frac{1}{2} O_2 \rightarrow \frac{1}{2} N_2O$
5. $\frac{1}{2} HgO \rightarrow \frac{1}{2} Hg + \frac{1}{2} O_2$
6. $\frac{1}{6} CO_2 + \frac{1}{6} H_2O \rightarrow \frac{1}{2} C_6H_{12}O_6 + \frac{1}{6} O_2$
7. $\frac{1}{2} HCI \rightarrow \frac{1}{2} CC_1^2 + \frac{1}{2} H_2$
8. $\frac{1}{3} SiCI_4 + \frac{1}{4} H_2O \rightarrow \frac{1}{2} NaOH + \frac{1}{4} HCI$
9. $\frac{1}{2} Na + \frac{1}{2} H_2O \rightarrow \frac{1}{2} NaOH + \frac{1}{4} HCI$
10. $\frac{1}{2} H_3PO_4 \rightarrow \frac{1}{4} H_2PO_7 + \frac{1}{4} H_2O$
11. $\frac{1}{2} CO_2 + \frac{1}{2} NH_3 \rightarrow \frac{1}{2} OC(NH_2)_2 + \frac{1}{2} H_2O$
13. $\frac{1}{4} Si_2H_3 + \frac{1}{7} O_2 \rightarrow \frac{1}{2} SiO_2 + \frac{1}{6} H_2O_3$
14. $\frac{1}{2} AI(OH)_3 + \frac{3}{3} H_2SO_4 \rightarrow \frac{1}{2} AI_2(SO_4)_3 + \frac{1}{6} H_2O$
15. $\frac{1}{4} Fe + \frac{3}{3} O_2 \rightarrow \frac{1}{2} Fe_2O_3$
16. $\frac{1}{4} Fe_2(SO_4)_3 + \frac{1}{6} KOH \rightarrow \frac{3}{3} K_2SO_4 + \frac{1}{2} Fe(OH)_3$
17. $\frac{1}{2} C_7H_6O_2 + \frac{1}{2} SO_2 \rightarrow \frac{1}{4} CO_2 + \frac{1}{6} H_2O$
18. $\frac{1}{4} SSO_4 + \frac{1}{2} HI \rightarrow \frac{1}{4} H_2S + \frac{4}{4} I_2 + \frac{4}{4} H_2O$
19. $\frac{1}{4} FeS_2 + \frac{11}{4} O_2 \rightarrow \frac{1}{4} Fe_2O_3 + \frac{1}{4} SO_2$
20. $\frac{1}{4} AI + \frac{3}{3} FeO \rightarrow \frac{1}{4} AI_2O_3 + \frac{3}{3} Fe$
21. $\frac{1}{4} Fe_2O_3 + \frac{3}{4} H_2 \rightarrow \frac{1}{4} Fe + \frac{3}{4} H_2O$
NayCO2 + $\frac{1}{4} HCI \rightarrow \frac{1}{4} NaCI + \frac{1}{4} H_2O$

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