

Study Guide Unit 9

1. KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):

a. Solution-

m. Solute-

b. Homogenous-

n. Solvent-

c. Heterogeneous-

o. Acid-

d. Soluble-

p. Base-

e. Insoluble-

q. Titration-

f. Miscible-

r. Analyte-

g. Immiscible-

s. Titrant-

h. Solubility-

t. Monoprotic acid-

i. Alloy-

u. Diprotic Acid

j. Saturated-

v. Triprotic Acid

k. Unsaturated-

l. Supersaturated-

2. How is the solubility of a gas affected by temp. and pressure?

Pressure up = solubility up

Temp up = solubility down

3. Under what conditions are salts most soluble?

High temperatures

4. Use the solubility curve to the right to answer the following.

a. What mass of KNO_3 will make a saturated solution at 60°C ? 100g

b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases?

NH_3 , $\text{Ce}_2(\text{SO}_4)_3$

c. What type of solution do I have if I **add** 110 g of NaNO_3 in 100g of 20°C water?

Saturated

d. What mass of KCl can go into solution at 80°C ?

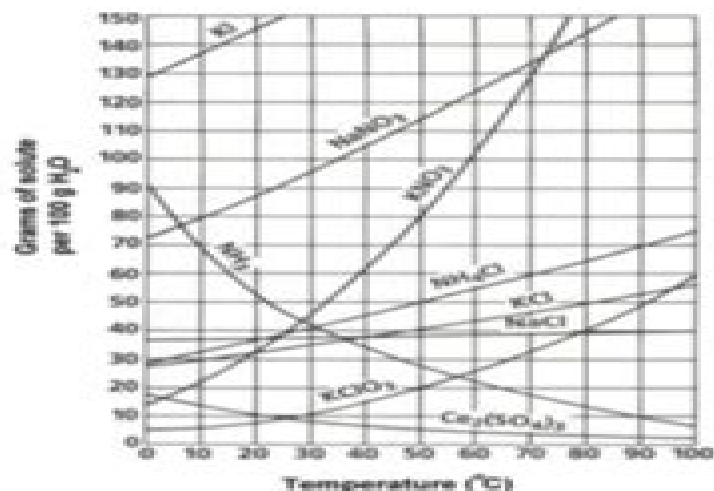
50 g

e. What is the least soluble compound at 20°C ?

KClO_3

f. Excluding KI , what is the most soluble compound at 80°C ?

KNO_3



Chem Solutions Study Guide Answers

Rachel S Tattersall



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Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

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