

Diagnostische toets

H10 Organische chemie

Deze toets bestaat uit 12 vragen.

Bij deze toets kun je maximaal 25 punten behalen.

Voor elk vraagnummer zijn de te behalen punten aangegeven.

Sorbinezuur

Sorbinezuur wordt gebruikt als conserveringsmiddel. Er wordt jaarlijks 30 000 ton van gefabriceerd. Voor de synthese van sorbinezuur worden ethenon en *trans*-but-2-eenol als grondstof gebruikt.



- 2p **1** Geef de systematische naam van sorbinezuur. Je hoeft hierbij geen rekening te houden met stereochemie.
- 3p **2** Geef de reactievergelijking in structuurformules van de reactie tussen ethenon en *trans*-but-2-eenol tot sorbinezuur.
- 1p **3** Tot welk type van organische reacties behoort deze vorming van sorbinezuur?

De oplosbaarheid van sorbinezuur in water is matig. Daarom wordt vaak gebruikgemaakt van een oplossing van een sorbinezuurzout. Deze oplossing wordt gemaakt door sorbinezuur, C_6H_7COOH , te mengen met natronloog. Hierbij treedt een reactie op waarbij een zoutoplossing ontstaat.

- 2p **4** Geef de reactievergelijking van sorbinezuur met natronloog in molecuulformules.

Het additiepolymeer van sorbinezuur wordt regelmatig gebruikt in crème. Polysorbinezuur kan goed vocht vasthouden, is vettig en smeerbaar. De additiepolymerisatie vindt alleen plaats op koolstofatomen 2 en 3 van het sorbinezuurmolecuul.

- 3p **5** Teken een stukje uit het midden van polysorbinezuur, waarin drie sorbinezuurmonomeren zijn weergegeven.
- 1p **6** Leg uit of sorbinezuur ook een condensatiepolymeer kan vormen.

Diagnostische Toets Pulsar Chemie

**Ludo Pieren, Rini Bekkers, Hielko de
Haan, Wijnand Rietman**



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