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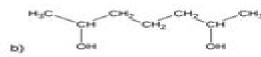
- 1. Draw the following alcohols
 - a) heptan-2-of
 - b) 3-methylhexan-1-oli
 - cyclopropanol

- 2,4,6-trichlorooctan-2-ol
- pentan-1,4-diol 60
- cyclohexene-1,3-diol

- 90 but-2-ene-1-of
- 4-methylpent-2-yne-1-ol-DO:

Name the following alcohols







3.4-dimethylcycloheptan-1-ol

Explain why the propane that is used as a fuel in a BBQ is gas at room temperature, but 2-propanol used as rubbing alcohol is a liquid at room temperature.

- 4. Draw the following ethers
 - a) 1-propoxypentane
 - b) 2-ethoxybutane

- 1-methoxy-4-chlorohexane
- 3-butoxy-2,4-dimethyloctane
- 2-propoxyheptane
- 1-ethoxycyclopentane

Name the following ethers:

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