

Balancing Equations and Simple Stoichiometry-KEY

Balance the following equations:

- 1) $1 \text{ N}_2 + 3 \text{ F}_2 \rightarrow 2 \text{ NF}_3$
- 2) $2 \text{ C}_6\text{H}_{10} + 17 \text{ O}_2 \rightarrow 12 \text{ CO}_2 + 10 \text{ H}_2\text{O}$
- 3) $1 \text{ HBr} + 1 \text{ KHCO}_3 \rightarrow 1 \text{ H}_2\text{O} + 1 \text{ KBr} + 1 \text{ CO}_2$
- 4) $2 \text{ GaBr}_3 + 3 \text{ Na}_2\text{SO}_3 \rightarrow 1 \text{ Ga}_2(\text{SO}_3)_3 + 6 \text{ NaBr}$
- 5) $3 \text{ SnO} + 2 \text{ NF}_3 \rightarrow 3 \text{ SnF}_2 + 1 \text{ N}_2\text{O}_3$

Using the following equation:



- 6) How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

$$? \text{ g Na}_2\text{SO}_4 = 200 \text{ g NaOH} \rightarrow \text{mol NaOH} \rightarrow \text{mol Na}_2\text{SO}_4 \rightarrow \text{g Na}_2\text{SO}_4$$

$$200 \text{ g NaOH} \times \frac{1 \text{ mol NaOH}}{40.00 \text{ g NaOH}} = 5 \text{ mol NaOH} \times \frac{1 \text{ mol Na}_2\text{SO}_4}{2 \text{ mol NaOH}} = 2.5 \text{ mol Na}_2\text{SO}_4$$

$$2.5 \text{ mol Na}_2\text{SO}_4 \times \frac{142 \text{ g Na}_2\text{SO}_4}{1 \text{ mol Na}_2\text{SO}_4} = \boxed{355 \text{ g Na}_2\text{SO}_4}$$

- 7) Using the following equation:



How many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate, assuming that you have an adequate amount of lead (IV) sulfate to do the reaction?

Molar masses: $\text{LiNO}_3 = 68.95 \text{ g/mol}$ & $\text{Li}_2\text{SO}_4 = 109.9 \text{ g/mol}$

$$? \text{ g LiNO}_3 = 250 \text{ g Li}_2\text{SO}_4 \rightarrow \text{mol Li}_2\text{SO}_4 \rightarrow \text{mol LiNO}_3 \rightarrow \text{g LiNO}_3$$

$$250 \text{ g Li}_2\text{SO}_4 \times \frac{1 \text{ mol Li}_2\text{SO}_4}{109.9 \text{ g Li}_2\text{SO}_4} = 2.27 \text{ mol Li}_2\text{SO}_4 \times \frac{4 \text{ mol LiNO}_3}{2 \text{ mol Li}_2\text{SO}_4} = 4.54 \text{ mol LiNO}_3$$

$$4.55 \text{ mol LiNO}_3 \times \frac{68.95 \text{ g LiNO}_3}{1 \text{ mol LiNO}_3} = \boxed{314 \text{ g LiNO}_3}$$

Chemfiesta Stoichiometry Lab Answers

Daniel F McAuley



Chemfiesta Stoichiometry Lab Answers:

Chemfiesta Stoichiometry Lab Answers Book Review: Unveiling the Power of Words

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Table of Contents Chemfiesta Stoichiometry Lab Answers

1. Understanding the eBook Chemfiesta Stoichiometry Lab Answers
 - The Rise of Digital Reading Chemfiesta Stoichiometry Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemfiesta Stoichiometry Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemfiesta Stoichiometry Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemfiesta Stoichiometry Lab Answers
 - Personalized Recommendations
 - Chemfiesta Stoichiometry Lab Answers User Reviews and Ratings
 - Chemfiesta Stoichiometry Lab Answers and Bestseller Lists

5. Accessing Chemfiesta Stoichiometry Lab Answers Free and Paid eBooks
 - Chemfiesta Stoichiometry Lab Answers Public Domain eBooks
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 - Chemfiesta Stoichiometry Lab Answers Budget-Friendly Options
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 - Chemfiesta Stoichiometry Lab Answers Compatibility with Devices
 - Chemfiesta Stoichiometry Lab Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemfiesta Stoichiometry Lab Answers
 - Highlighting and Note-Taking Chemfiesta Stoichiometry Lab Answers
 - Interactive Elements Chemfiesta Stoichiometry Lab Answers
8. Staying Engaged with Chemfiesta Stoichiometry Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemfiesta Stoichiometry Lab Answers
9. Balancing eBooks and Physical Books Chemfiesta Stoichiometry Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemfiesta Stoichiometry Lab Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemfiesta Stoichiometry Lab Answers
 - Setting Reading Goals Chemfiesta Stoichiometry Lab Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemfiesta Stoichiometry Lab Answers
 - Fact-Checking eBook Content of Chemfiesta Stoichiometry Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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