MULTIPLE CHOICE QUESTIONS (MCQ)

		THE CALL BUILD CORE	OICE QUEST	10113 (11102)
1.	Which of the following is dependent of temperature?			
	(a)	Molarity	(b)	Molality
	(c)	Mole fration	(d)	Mass percentage
2.	4 g of NaOH dissolved in 100 ml solution. Molarity of the solution is			
	(a)	1 M	(b)	10 M
	(c)	0.1 M	(d)	4 M
3.	Which has the maximum number of molecules among the following?			
	(a)	44g of CO ₂	(b)	44g of 0.
	(c)	8g of H ₂	(d)	64g of SO ₂
4.	10 mol of Zn react with 10 mol of HCl. Calculate the number of moles of H ₂ produced.			
	(a)	5 mol	(6)	10 mol
	(c)	20 mol	(p)	2.5 mol
5.	The number of oxygen atoms in 4.4g of CO ₂ is approximately			
	(a)	1.2 × 10 ²³	(b)	6×10^{22}
	(c)	6 × 10 ²³	(d)	12×10^{23}
6.	The molarity of a solution obtained by mixing 750 mL of 0.5 M HCl with 250 ml of 2 M HCl will be			
	(a)	0.975 M	(b)	0.875 M
	(c)	1-00 M	(b)	1.175 M
7.	Number of atoms of He in 100 u of He (Atomic mass of He is 4 u)			
	(a)	25	(b)	50
	(c)	100	(d)	400
8.	6.02×10^{20} molecules of urea are present in 100 mL of its solution. The concentration of the solution is			
	(a)	0.02 M	(b)	0.01 M
	(c)	0.001 M	(d)	0.1 M

Chem Mcq Question In Hs

Léo Pomar

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