

Post-Lab Questions

1. From the experiment, which reactant is the limiting reactant? Which one is the reactant in excess? How do you determine which one is the limiting reactant?
2. If you weigh the watch glass then write your name on it with the wax pencil, how will that affect your percent yield?
3. What factors could cause the percent yield of a product to be less than 100%?
4. What factors could cause the percent yield of a product to be more than 100%?
5. Balance the following reaction: $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 - a. 4.30 g of C_3H_8 reacts with 4.30 g of O_2 . Which one is the limiting reagent, and which one is in excess? Explain your reasoning.
 - b. Based on the limiting reagent, what are the theoretical yields of CO_2 and H_2O ?
 - c. You perform the experiment and produce 2.67 g of CO_2 and 0.79 g of H_2O . What are your percent yields for each product?

Chem 1211 Lab Manual Answers

JA Banks



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Chem 1211 Lab Manual Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Chem 1211 Lab Manual Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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