

Chemistry Bonding MCQ (From 2015 Prelim Papers)

Q1

In which of the following pairs does the first species have a larger bond angle than the second?

- A SO_2 , PI_3
- B SO_3^{2-} , SO_4^{2-}
- C H_2S , NH_3
- D H_2O , H_3O^+

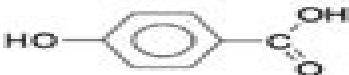
Q2

Which of the following statements is correct?

- A BCl_3 and BrCl_3 have the same molecular shape.
- B Cyclohexene and propanone are non-polar liquids.
- C LiAlH_4 and NaBF_4 contain dative covalent bonds.
- D XeF_4 and $\text{H}_2\text{C}=\text{C}=\text{CH}_2$ are planar molecules.

Q3

In which of the following pairs does the first substance have a higher melting point than the second?

	First substance	Second substance
A	$\text{CH}_3\text{CH}_2\text{OCH}_3$	$\text{CH}_3\text{CH}_2\text{NHCH}_3$
B	$\text{CH}_3(\text{CH}_2)_4\text{CH}_3$	$(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)_2$
C	RbCl	KCl
D		

Q4

Which of the following is a planar molecule?

- A $\text{C}_2\text{F}_4^{2+}$
- B $\text{CH}_2=\text{C}=\text{CH}_2$
- C 
- D $\text{CH}_2=\text{CH}-\text{CHO}$

Chemical Bonding Multiple Choice Test

Annelies Wilder-Smith



Chemical Bonding Multiple Choice Test:

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Table of Contents Chemical Bonding Multiple Choice Test

1. Understanding the eBook Chemical Bonding Multiple Choice Test
 - The Rise of Digital Reading Chemical Bonding Multiple Choice Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bonding Multiple Choice Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Chemical Bonding Multiple Choice Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bonding Multiple Choice Test
 - Personalized Recommendations
 - Chemical Bonding Multiple Choice Test User Reviews and Ratings

- Chemical Bonding Multiple Choice Test and Bestseller Lists
- 5. Accessing Chemical Bonding Multiple Choice Test Free and Paid eBooks
 - Chemical Bonding Multiple Choice Test Public Domain eBooks
 - Chemical Bonding Multiple Choice Test eBook Subscription Services
 - Chemical Bonding Multiple Choice Test Budget-Friendly Options
- 6. Navigating Chemical Bonding Multiple Choice Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Bonding Multiple Choice Test Compatibility with Devices
 - Chemical Bonding Multiple Choice Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Bonding Multiple Choice Test
 - Highlighting and Note-Taking Chemical Bonding Multiple Choice Test
 - Interactive Elements Chemical Bonding Multiple Choice Test
- 8. Staying Engaged with Chemical Bonding Multiple Choice Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Bonding Multiple Choice Test
- 9. Balancing eBooks and Physical Books Chemical Bonding Multiple Choice Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Bonding Multiple Choice Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Bonding Multiple Choice Test
 - Setting Reading Goals Chemical Bonding Multiple Choice Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Bonding Multiple Choice Test
 - Fact-Checking eBook Content of Chemical Bonding Multiple Choice Test
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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