

Chapter 19

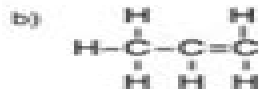
1. a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carbon-carbon single bonds.



(ii) liquid



2. a) a carbon-carbon double bond



(Home schooling parents: It doesn't matter what position you draw the bromine in on the right-hand carbon - N, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)



(ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

3. a) Ethene is passed with steam at 300°C and 60 - 70 atmospheres over a phosphoric acid catalyst.

b) e.g. two of: continuous flow process, therefore faster; rapid reaction; purer product.

c) uses renewal resources; gentle conditions.

4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out while allowing carbon dioxide to escape. You could use a commercial home-winemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.



Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at 78°C . (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

continued . . .

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Victor M. Corman



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Table of Contents Catalyst Lab Manual Chem 101 Answer Solutions

1. Understanding the eBook Catalyst Lab Manual Chem 101 Answer Solutions
 - The Rise of Digital Reading Catalyst Lab Manual Chem 101 Answer Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Catalyst Lab Manual Chem 101 Answer Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Catalyst Lab Manual Chem 101 Answer Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Catalyst Lab Manual Chem 101 Answer Solutions
 - Personalized Recommendations

- Catalyst Lab Manual Chem 101 Answer Solutions User Reviews and Ratings
- Catalyst Lab Manual Chem 101 Answer Solutions and Bestseller Lists
- 5. Accessing Catalyst Lab Manual Chem 101 Answer Solutions Free and Paid eBooks
 - Catalyst Lab Manual Chem 101 Answer Solutions Public Domain eBooks
 - Catalyst Lab Manual Chem 101 Answer Solutions eBook Subscription Services
 - Catalyst Lab Manual Chem 101 Answer Solutions Budget-Friendly Options
- 6. Navigating Catalyst Lab Manual Chem 101 Answer Solutions eBook Formats
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 - Catalyst Lab Manual Chem 101 Answer Solutions Compatibility with Devices
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 - Highlighting and Note-Taking Catalyst Lab Manual Chem 101 Answer Solutions
 - Interactive Elements Catalyst Lab Manual Chem 101 Answer Solutions
- 8. Staying Engaged with Catalyst Lab Manual Chem 101 Answer Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Catalyst Lab Manual Chem 101 Answer Solutions
- 9. Balancing eBooks and Physical Books Catalyst Lab Manual Chem 101 Answer Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Catalyst Lab Manual Chem 101 Answer Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Catalyst Lab Manual Chem 101 Answer Solutions
 - Setting Reading Goals Catalyst Lab Manual Chem 101 Answer Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Catalyst Lab Manual Chem 101 Answer Solutions
 - Fact-Checking eBook Content of Catalyst Lab Manual Chem 101 Answer Solutions

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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