

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 3

Q1. In a hydrocarbon, the mass ratio of hydrogen and carbon is 1:3; the empirical formula of hydrocarbon is ____.

- a.) CH_4
- b.) CH_2
- c.) C_2H
- d.) CH_3

Q2. 80g of oxygen contains as many atoms as in ____.

- a.) 10 g of hydrogen
- b.) 5 g of hydrogen
- c.) 8 g of hydrogen
- d.) 1 g of hydrogen

Q3. A mixture of gases contains H_2 and O_2 gases in the ratio 1:4. What is the molar ratio of the two gases in the mixture?

- a.) 16:1
- b.) 2:1
- c.) 1:4
- d.) 4:1

Q4. Which of the following concentration terms is/are independent of temperature?

- a.) Molality
- b.) Mole fraction
- c.) Molarity
- d.) Normality

Q5. Which of the following contains the highest number of atoms?

- a.) 1.0 g of water
- b.) 1.0 g of silver
- c.) 1.0 g of nitrogen
- d.) 1.0 g of propane

Chemistry Chapter 11 Guided Reading Answers

TD Snyder



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Reviewing **Chemistry Chapter 11 Guided Reading Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemistry Chapter 11 Guided Reading Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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