

Chemistry 100 Final Study Guide

- **Chapter 1: Introduction**

- a. Physical vs. Chemical properties
 - 1) List 2 physical and 1 chemical property of magnesium.
- b. Units of measurement
 - 1) Can you distinguish between metric and U.S. units of measurement?
 - 2) What is 0.000000568 in scientific notation? What is 4.63×10^7 in regular notation?
- c. Density
 - 1) What is the density of a 37g block of metal that takes up 15mL of volume?

- **Chapter 3: Atomic Structure**

- a. Historical Perspective
 - 1) List the major scientific accomplishments of the following scientists:
 - (i) Michael Faraday
 - (ii) Joseph Thomson
 - (iii) Robert Millikan
 - (iv) Ernest Rutherford
 - (v) Neils Bohr
 - (vi) Erwin Schrodinger
- b. Isotopes
 - 1) Define an isotope and give an example of several isotopes for hydrogen.
- c. Structure of the atom
 - 1) Name the 3 main parts of an atom, and indicate where in the atom they'd be found.
 - 2) For the element sodium: how many energy levels? How many sublevels? What orbitals? How many electrons?
 - 3) Write the full electron configuration of silicon.
- d. Electron configurations and trends on the Periodic Table
 - 1) Locate each of the following on the Periodic Table: alkali metals, alkaline earth metals, transition metals, halogens, noble gases, metals, non-metals.

- **Chapter 5: Chemical bonds**

- a. Electron-dot structures
 - 1) Draw the electron-dot structure of an oxygen atom.
- b. Octet Rule and Ions
 - 1) What charge of ions do magnesium, chlorine and nitrogen form?
- c. Ionic compounds
 - 1) Name the following compounds: AlBr_3 , CaS , Na_3N
- d. Covalent compounds
 - 1) Name the following compounds: SO_2 , PCl_3 , CH_4
- e. Polyatomic molecules
 - 1) Name and write the formula for the 6 -ate polyatomic molecules we discussed.
- f. Electron-dot formulas for molecules
 - 1) Write the electron-dot formula for each of the 6 -ate polyatomic ions we discussed.
- g. VSEPR Model and Molecular shape
 - 1) Draw the electron-dot formula of ammonia (NH_3) and state its VSEPR shape.
- h. Molecule polarity
 - 1) Which of the VSEPR molecule shapes are polar and which are non-polar?

Chemistry Final Exam Study Guide 100 Questions

Daniel F McAuley



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Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26

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A Science Career Against all Odds Bernhard Wunderlich, 2010-03-01 Today is Sunday June 17 2007 Father s Day Naturally the obligatory carefully selected cards phone calls and small gifts arrived from the children and grandchildren Best wishes for Father s Day were also the first words in the morning from Heidi my wife of 54 years although for many years I had made the comment I am not your father But in the frame of my life s experiences th th in the 20 century as I intend to summarize them over the next few years the 17 of June has much deeper significance This was the day in 1953 when we finally fled from our life of oppression which had lasted 20 years Two successive dictatorships one of Hitler and the other of Stalin caused the most horrific slaughter of civilians and soldiers eclipsing all prior history During these first years of my life I was plainly lucky to survive After this day I had a much better chance to experience the freedom needed to lead a life of creativity satisfaction and ultimately prosperity all directed largely by our own decisions th The 17 of June 1953 was a Wednesday I stayed in the apartment of my parents in my hometown of Brandenburg in the German Democratic Republic GDR the former Russian occupied zone of Germany The summer vacation of the Humboldt University in East Berlin some 40 mi further east had just started But I was alone with my father Vati

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