

**CP Chemistry - Zajac - Semester II Exam Study Guide / Practice Exam****Multiple Choice***Identify the choice that best completes the statement or answers the question.*

- \_\_\_\_\_ 1. All of the following are equal to Avogadro's number EXCEPT \_\_\_\_\_.  
a. the number of atoms of bromine in 1 mol  $\text{Br}_2$   
b. the number of atoms of gold in 1 mol Au  
c. the number of molecules of nitrogen in 1 mol  $\text{N}_2$   
d. the number of molecules of carbon monoxide in 1 mol CO
- \_\_\_\_\_ 2. How many moles of tungsten atoms are in  $4.8 \times 10^{23}$  atoms of tungsten?  
a.  $8.0 \times 10^2$  moles  
b.  $8.0 \times 10^3$  moles  
c.  $1.3 \times 10^{-1}$  moles  
d.  $1.3 \times 10^{-2}$  moles
- \_\_\_\_\_ 3. Which of the following is NOT a true about atomic mass?  
a. The atomic mass is 12 g for magnesium.  
b. The atomic mass is the mass of one mole of atoms.  
c. The atomic mass is found by checking the periodic table.  
d. The atomic mass is the number of grams of an element that is numerically equal to the mass in amu.
- \_\_\_\_\_ 4. The mass of a mole of NaCl is the \_\_\_\_\_.  
a. molar mass  
b. atomic mass  
c. molecular mass  
d. gram atomic mass
- \_\_\_\_\_ 5. What is the molar mass of  $\text{AuCl}_3$ ?  
a. 96 g  
b. 150 g  
c. 232.5 g  
d. 303.6 g
- \_\_\_\_\_ 6. The molar mass of  $\text{C}_3\text{H}_8$  and the molar mass of  $\text{CaCO}_3$  contain approximately the same number of \_\_\_\_\_.  
a. carbon atoms  
b. cations  
c. anions  
d. grams
- \_\_\_\_\_ 7. How many moles of  $\text{CaBr}_2$  are in 5.0 grams of  $\text{CaBr}_2$ ?  
a.  $2.5 \times 10^{-2}$  mol  
b.  $4.2 \times 10^{-2}$  mol  
c.  $4.0 \times 10^2$  mol  
d.  $1.0 \times 10^2$  mol
- \_\_\_\_\_ 8. The lowest whole-number ratio of the elements in a compound is called the \_\_\_\_\_.  
a. empirical formula  
b. molecular formula  
c. binary formula  
d. representative formula

# Chemistry Cp Study Guide Answers

**JoAnne Silbert-Flagg, Elizabeth D.  
Sloand**



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