

CALORIMETRY GUIDED NOTES

Lesson Duration: 60 Minutes

Description:

The intention behind these Calorimetry Guided Notes is to introduce students to transfer of heat between two substances. Students will answer scaffolded questions to guide their understanding and complete a guided practice problem prior to completing two of their own practice questions. This student-driven worksheet allows teachers to facilitate learning indirectly while students work in small groups or individually. This lesson builds off of specific heat lessons and can prepare students for a labs on calorimetry lab or unknown metal lab.

This product can easily be used as a print-out activity, or as a TpT Digital Activity.

This product includes an answer key!

Students will need a calculator to complete this activity.

Learning Standards:

Science Practices: HS-PS3-4 Energy: Plan and conduct an investigation to provide evidence that the transfer of thermal energy when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (science of thermodynamics).

Disciplinary Core Ideas: PS3-1 Conservation of Energy and Energy Transfer: Energy cannot be created or destroyed, but it can be transferred from one place to another, and transformed between systems.

Cross-Cutting Concept: When investigating or describing a system, the boundaries and initial conditions of the system need to be defined and their inputs and outputs analyzed and described using models.

Science and Engineering Practices: Plan and conduct an investigation individually to produce data to serve as the basis for evidence, and in the design, decide on types, how much, and accuracy of data needed to produce reliable measurements and consider limitations on the precision of the data and refine the design accordingly.

Learning Objectives:

Students will be able to...

- Differentiate between temperature and heat
- Calculate heat lost or gained within a system
- Isolate variables within an equation
- Identify an unknown metal based on the heat
- Solve for various unknowns within the lab

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Teacher Guide

Calorimetry Guided Activity Student Guide

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