#### CHEMISTRY TEST NAMES INTRODUCTORY CHEMISTRY CONCEPTS DATE: Directions: For each of the following questions, choose the number that best answers the question and place it on your answer sheet. Directions: For each of the following questions, all major steps must be shown to receive full credit. Introductory Chemistry Concepts Level 1 Multiple Choice A chemical change always results characteristically in a change of 10 states: 2) forme identity: 4) size: shape. ANS: 3 Which of the following is a chemical change? 1) evaporation of water 2) distillation of water 3) freezing of water 4) electrolysis of water ANS: 4 Chemical changes always involve 1) the absorption of energy; the liberation. either the absorption or liberation of energy; 4) changes of state. of energy: ANS: 3 One physical property of sugar is its 1) acidity: reactivity; 3) flammability: 4) plasma qualities: solubility. ANS: 5 The majority of chemical changes are I hautothermie: 2) endothermie: isothermic: 4) nonthermic: 5) exothermic. ANS: 5 6. The statement which describes a physical property of the element oxygen is oxygen supports the burning of paper; 2) oxygen is needed for the rusting of 3) oxygen must be present for food to decay; 4) oxygen has a density of 1.43 grams per liter; oxygen is a component of water. ANS: 4 In any chemical reaction energy is 1) always absorbed; always. 3) released as heat: 4) usually not involved: metassissed: 5) either absorbed or released. A.NS: 5 8. Which of the following is **not** a chemical change? 2) the exploding of gasoline I) the dissolving of zinc by HCl 3) the corroding of a copper roof 4) the boiling of water ANS: 4 9. One chemical property of a substance is a description of its 1) odor: density; combustibility; 4) solubility: 5) none of these. ANS: 3 2) chemical A new substance is formed as a result of a 1) physical change;

# **California Chemistry Matter And Change Answer Key**

**Lauren Gardner** 

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