

Preparation for Exam 2, November 18
CHM2046C, Spring 2022

1

1. For all calculations, remember to use proper units, and show your work! For possible partial credit, distinguish between partial and full credit questions in which it says "100% (Full Credit)".

a. Give the name and the abbreviations for each of the following systems of units:

ai) The SI system with its relation to the metric system: _____

bi) The SI system with its relation to the metric system: _____

2. Classify each of the following as an ionic compound, molecular, hydrogen-bonded, or hydrogen-bonded ionic liquid:

ai) Hydrogen chloride (HCl , H_2O , H_2O_2)

bi) Hydrogen chloride/ethanol

3. Identify each of the following as a physical (P) or a chemical (C) change:

ai) Ethanol is burned. P ☐ C ☐

bi) Ethanol is separated by a solid material, producing two parts. P ☐ C ☐

4. Characterize each of the following substances as to its state:

ai) _____, Hydrogen chloride is a substance of the following nature: is a substance

bi) _____, Air is most commonly a gas mixture

ci) _____, Liquid water is a substance and is a substance

5. Circle the correct temperature:

-1.3°C

0°F

2400K

6. Explain the following physical properties in terms of molecular structure:

Chem 139 Practice Tests

Sebastian Brünink



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Table of Contents Chem 139 Practice Tests

1. Understanding the eBook Chem 139 Practice Tests
 - The Rise of Digital Reading Chem 139 Practice Tests
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem 139 Practice Tests
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem 139 Practice Tests
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem 139 Practice Tests
 - Personalized Recommendations
 - Chem 139 Practice Tests User Reviews and Ratings

- Chem 139 Practice Tests and Bestseller Lists
- 5. Accessing Chem 139 Practice Tests Free and Paid eBooks
 - Chem 139 Practice Tests Public Domain eBooks
 - Chem 139 Practice Tests eBook Subscription Services
 - Chem 139 Practice Tests Budget-Friendly Options
- 6. Navigating Chem 139 Practice Tests eBook Formats
 - ePub, PDF, MOBI, and More
 - Chem 139 Practice Tests Compatibility with Devices
 - Chem 139 Practice Tests Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem 139 Practice Tests
 - Highlighting and Note-Taking Chem 139 Practice Tests
 - Interactive Elements Chem 139 Practice Tests
- 8. Staying Engaged with Chem 139 Practice Tests
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem 139 Practice Tests
- 9. Balancing eBooks and Physical Books Chem 139 Practice Tests
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem 139 Practice Tests
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chem 139 Practice Tests
 - Setting Reading Goals Chem 139 Practice Tests
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chem 139 Practice Tests
 - Fact-Checking eBook Content of Chem 139 Practice Tests
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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