Naming Ionic Compounds Practice Worksheet

silver bromide

20

Lots of Ionic Naming Practice Problems

Name the following ionic compounds:		Name the following io	Name the following ionic compounds:	
1)	NH ₄ CI	1) Naßr		
2)	Fe(NO ₃) ₃	2) So(OH) ₃		
3)	Ter,	3) V ₂ (SO ₄) ₅		
4)	Cu ₂ P	4) NH,F		
5)	SnSe ₉	 6) CaCO₃ 		
6)	GaAs	 NPO_x 		
7):	Pb(SO _u) _d	7) U ₀ SO ₃		
8)	Be(HCO ₁) _V	8) Zn ₀ P ₂		
90	Mng(SO _{sile}	9) Sr(C ₂ H ₂ O ₂) ₀		
10)	Al(CN) ₀	10) Cu ₂ O		
Write the formulas for the following compounds:		11) Ag ₂ PO ₄	17.4	
113	chromium (VI) phosphate	12) YOIO ₉		
12)	vanadium (IV) carbonate	13) SnS ₂		
130	Sin (III) nibrite	14) Ti(CN) ₄		
14)	cobalt (III) oxide	15) KMnO ₊		
15)	Stanium (II) acetate	16) Pb ₃ N ₂		
16)	vanadium (V) suffide	17) CoCO ₃	1	
17)	chromium (IIII) hydraxide	18) CdSO ₁		
18)	lithium iodide	19) Cu(NO ₂) ₂		
19)	lead (II) nitride	20) Fe(HCO ₃) ₂		

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Great Minds

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