

Shiloh Season by Phyllis Reynolds Naylor

Reading comprehension questions

1. What is Marty's primary goal in the story?
2. How does Judd Travers change throughout the book?
3. What role does Shiloh play in Marty's life?
4. Describe the relationship between Marty and his family.
5. How does Marty feel about confronting Judd?
6. What motivates Marty to keep Shiloh safe?
7. How does the setting impact the story's events?
8. Discuss the theme of forgiveness in the book.
9. What does Marty learn about responsibility?
10. How does Marty's perception of Judd change?
11. What conflicts does Marty face internally?
12. How does Marty's friendship with David change?
13. Explore the theme of trust in the book.
14. What role does Dana Lynn play in Marty's decisions?
15. Discuss the significance of the title, "Shiloh Season."
16. How does Marty's view of justice evolve?
17. What sacrifices does Marty make for Shiloh?
18. Analyze the dynamics between Marty and his father.
19. How does Marty's perspective on bravery shift?
20. What impact does Judd's backstory have on the narrative?
21. How does the community react to Marty's actions?
22. What role does Doc Murphy play in Marty's journey?
23. Analyze the symbolism behind Shiloh's name.
24. How does Marty's moral compass guide his choices?
25. Explore the themes of compassion and empathy in the story.
26. Discuss the parallels between Marty and Judd.
27. How does Marty's relationship with Shiloh parallel other relationships in the book?
28. What lessons does Marty learn about animal rights?
29. How does Marty deal with the ethical dilemmas he faces?
30. What does Marty's persistence reveal about his character?
31. How does Marty's viewpoint on second chances change?

Comprehension Questions For Shiloh Season

Michael Rajnik



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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely

divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalization of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7.0002g of copper ii sulfate and 2.0101g of iron metal which produced 2.3037g of copper product with a percent yield of 100.72 introduction

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