

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	18
_____	Hydrogen
_____	Li (103)
_____	Neon (1018)

Atomic Mass	19.0
Atomic Number	18.0
Electron Configuration	1S ²
Chemical Name	Ne
Chemical Symbol	Ne

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. There are two periods of the elements in the periodic table.
- _____ 26. Group 1 elements except for hydrogen and lithium are the alkali metals.
- _____ 27. Group 18 elements are the noble gases.
- _____ 28. Group 1 elements are highly reactive elements in the periodic table.
- _____ 29. Group 18 elements are very reactive elements in the periodic table.
- _____ 30. Elements have properties of both metals and nonmetals.

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Gregory J. Privitera



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Table of Contents Content Mastery Study Guide Key

1. Understanding the eBook Content Mastery Study Guide Key
 - The Rise of Digital Reading Content Mastery Study Guide Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Content Mastery Study Guide Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Content Mastery Study Guide Key
 - User-Friendly Interface

4. Exploring eBook Recommendations from Content Mastery Study Guide Key
 - Personalized Recommendations
 - Content Mastery Study Guide Key User Reviews and Ratings
 - Content Mastery Study Guide Key and Bestseller Lists
5. Accessing Content Mastery Study Guide Key Free and Paid eBooks
 - Content Mastery Study Guide Key Public Domain eBooks
 - Content Mastery Study Guide Key eBook Subscription Services
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6. Navigating Content Mastery Study Guide Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Content Mastery Study Guide Key Compatibility with Devices
 - Content Mastery Study Guide Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Content Mastery Study Guide Key
 - Highlighting and Note-Taking Content Mastery Study Guide Key
 - Interactive Elements Content Mastery Study Guide Key
8. Staying Engaged with Content Mastery Study Guide Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Content Mastery Study Guide Key
9. Balancing eBooks and Physical Books Content Mastery Study Guide Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Content Mastery Study Guide Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Content Mastery Study Guide Key
 - Setting Reading Goals Content Mastery Study Guide Key
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Content Mastery Study Guide Key
 - Fact-Checking eBook Content of Content Mastery Study Guide Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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pb no3 2 aq 2 hcl aq pbcl2 s 2 hno3

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