Using the first molecule as a guide, in Model 12.1 write each molecule's:

Model 12.1: Constitutional isomers

Constitutional (also known as structural) isomers have the same chemical formula but different atom connectivity. The following hydrocarbons are all constitutional isomers:

Line structure	Name	Functional group	Chemical formula
	1-butene	Alkene	C_4H_8
_			
<u>`</u>			
人			

a. Name 2-butene 2-butene 2-methyl-1-propene Cyclobutane 1-methylcylopropane

Cycloalkane

- b. Functional group, and:
 Alkene
 Alkene
 Cycloalkane
- c. Chemical formula
 C₁H₃
 C₁H₃
 C₁H₃
 C₁H₃
 C₁H₃
- Based on the information determined in Question 1, which features are sometimes different for each of the molecules above?

The name, shape and functional group

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